

Examples of multiple choice math exam questions for grade 3 high school along with answer keys and explanations

1. If  $f(x) = 3x + 2$  and  $g(x) = 2x - 5$ , what is the value of  $f(g(4))$ ?

a. 5  
b. 11  
c. 14  
d. 23

Answer: c. 14. Solution: First, we need to evaluate  $g(4)$  which gives  $2(4) - 5 = 3$ . Then, we evaluate  $f(g(4))$  which gives  $f(3) = 3(3) + 2 = 11$ .

2. What is the equation of a line perpendicular to  $y = 3x - 2$  and passing through the point  $(2, 4)$ ?

a.  $y = -1/3x + 10/3$   
b.  $y = -3x + 10$   
c.  $y = 1/3x + 10/3$   
d.  $y = 3x - 10$

Answer: a.  $y = -1/3x + 10/3$ . Solution: The slope of a line perpendicular to  $y = 3x - 2$  is  $-1/3$ , because the product of the slopes of perpendicular lines is  $-1$ . Using the point-slope form of the equation of a line gives  $y - 4 = (-1/3)(x - 2)$ , which can be simplified to  $y = -1/3x + 10/3$ .

3. If a function is defined as  $f(x) = x^2 - 4x$ , what is the value of  $f(-2)$ ?

a. 4  
b. 8  
c. 12  
d. 16

Answer: d. 16. Solution: Substituting  $x = -2$  into the function gives  $f(-2) = (-2)^2 - 4(-2) = 16$ .

4. What is the value of  $\tan(45^\circ) + \cos(60^\circ)$ ?

a.  $2 + \sqrt{3}$   
b.  $\sqrt{2} + \sqrt{3}$   
c.  $2\sqrt{2} + 1$   
d.  $\sqrt{3} + 1$

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**D Kirk**



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web the amount of  $\text{KNO}_3$  in the solution is called solubility if the  $\text{KNO}_3$  is less than what the water can handle then it is an unsaturated solution if the amount of  $\text{KNO}_3$  is equal to

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potassium nitrate has an orthorhombic crystal structure at room temperature which transforms to a trigonal system at 128 c 262 f on cooling from 200 c 392 f another trigonal phase forms between 124 c 255 f and 100 c 212 f sodium nitrate is isomorphous with calcite the most stable form of calcium carbonate whereas room temperature potassium nitrate is isomorphous with aragonite a slightly less stable polymorph of

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web a solution with the maximum possible amount of solute is saturated if a solution contains less than the maximum amount of solute it is unsaturated when a solution is

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web a saturated solution is a solution that contains the maximum amount of solute that is capable of dissolving an unsaturated solution is a solution that contains less than the

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web nov 13 2022 the important point to remember is that any conversion involving molarity requires a knowledge of the density of the solution example pageindex 9 a

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web in the case of a saturated solution of  $\text{KNO}_3$  when it is cooled from 80 degrees to 50 degrees some of the  $\text{KNO}_3$  is expected to crystallize out of the solution the amount of

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