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**EQUIPMENT DIVISION**

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**TECHNICAL MANUAL**



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North American Version  
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# Jd Lt160 Manual

**C Cleary**



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base than  $\text{Cl}^-$   $\text{pK}_a$  7.7 to identify the

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web acids and bases section 1 answer the following questions in the space provided a  $\text{H}_2\text{SO}_4$  b  $\text{H}_2\text{SO}_3$  c  $\text{H}_2\text{S}$  d  $\text{HClO}_4$  e hydrogen cyanide which if any of the acids

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web characteristics physical acids would look solid liquid or in the form of gas it would also have a sour taste bases would feel slippery and solid in nature except for ammonia

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web log  $\text{H}^+$  10<sup>-4</sup> 0.5 8.91  $\times 10^{-5}$  mol dm<sup>-3</sup> log  $\text{pK}_a$   $\text{pH}$  3.75 4.05 0.30 10.0 3.0 0.50 and  $\text{HCOOH}$  0.1 so  $\text{HCOO}^-$  0.200 mol dm<sup>-3</sup> amount vol x

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web only the better candidates could write successive ionisation equations in part c while part d about the  $\text{pH}$  of equimolar strong and weak acid solutions was poorly answered

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web a substance may be assigned to one our four conceivable categories it may be an acid or a base but in addition it may be both an acid and a base or it may be neither an acid nor

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web to detect the presence of acids and bases micheal adds a few drops of phenolphthalein indicator to two colorless solutions a and b he observes the following changes in the

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web nov 5 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low  $\text{pH}$

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web section 2  $\text{pH}$  of strong acids number of protons released monoprotic acid acid that releases one  $\text{H}^+$  ion per molecule e.g  $\text{HCl}$  hydrochloric acid  $\text{HNO}_3$  nitric acid  $\text{CH}_3\text{COOH}$

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web jan 10 2021  $\text{S}^{2-}$  3.21 strong electrolytes are 100 ionized and as long as the component ions are neither weak acids nor weak bases the ionic species present result

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web problem 1 which of the following are not acids  $\text{CH}_3\text{COOH}$   $\text{CO}_2$   $\text{HNO}_2$   $\text{HCOOH}$   $\text{CCl}_4$   $\text{CH}_3\text{COOH}$   $\text{CO}_2$   $\text{HNO}_2$   $\text{HCOOH}$   $\text{CH}_3\text{COOH}$

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web in grade 11 learners were introduced to a lot of the concepts that are expanded on in this chapter these include acid and base models and definitions conjugate acid base

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web which is the conjugate base in the following reaction multiple choice question 13 which of the following compounds is the strongest acid multiple choice question 14 which

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web 11 why is hcl a strong acid and hclo a weak acid 12 why are hcl and hclo 4 both strong acids 13 for each of the reactions below classify the reactants as an acid or a

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