

JEE Advanced 2020 | Paper-2 | Code-E

4. An acidified solution of potassium chromate was layered with an equal volume of amyl alcohol. When it was shaken after the addition of 1 mL of 3% H_2O_2 , a blue alcohol layer was obtained. The blue color is due to the formation of a chromium (VI) compound 'X'. What is the number of oxygen atoms bonded to chromium through only single bonds in a molecule of X?

Answer: 4

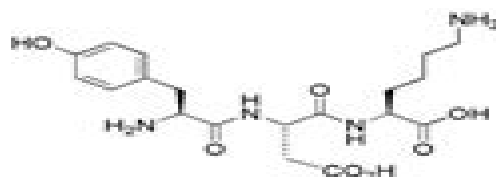
Solution:

Blue colour compound of 'Cr' is CrO_5 .



Oxygen atoms bonded to chromium through only single bonds = 4

5. The structure of a peptide is given below.



If the absolute values of the net charge of the peptide at $\text{pH} = 2$, $\text{pH} = 6$, and $\text{pH} = 11$ are $|z_1|$, $|z_2|$, and $|z_3|$, respectively, then what is $|z_1| + |z_2| + |z_3|$?

Answer: 5

Solution:

(i) At $\text{pH} = 2$ (Highly acidic)

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Patrick Vollmar



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Table of Contents Jee Exam Paper 2 Code K Solution

1. Understanding the eBook Jee Exam Paper 2 Code K Solution
 - The Rise of Digital Reading Jee Exam Paper 2 Code K Solution
 - Advantages of eBooks Over Traditional Books
2. Identifying Jee Exam Paper 2 Code K Solution
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jee Exam Paper 2 Code K Solution
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jee Exam Paper 2 Code K Solution
 - Personalized Recommendations
 - Jee Exam Paper 2 Code K Solution User Reviews and Ratings
 - Jee Exam Paper 2 Code K Solution and Bestseller Lists
5. Accessing Jee Exam Paper 2 Code K Solution Free and Paid eBooks

- Jee Exam Paper 2 Code K Solution Public Domain eBooks
- Jee Exam Paper 2 Code K Solution eBook Subscription Services
- Jee Exam Paper 2 Code K Solution Budget-Friendly Options
- 6. Navigating Jee Exam Paper 2 Code K Solution eBook Formats
 - ePub, PDF, MOBI, and More
 - Jee Exam Paper 2 Code K Solution Compatibility with Devices
 - Jee Exam Paper 2 Code K Solution Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jee Exam Paper 2 Code K Solution
 - Highlighting and Note-Taking Jee Exam Paper 2 Code K Solution
 - Interactive Elements Jee Exam Paper 2 Code K Solution
- 8. Staying Engaged with Jee Exam Paper 2 Code K Solution
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jee Exam Paper 2 Code K Solution
- 9. Balancing eBooks and Physical Books Jee Exam Paper 2 Code K Solution
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jee Exam Paper 2 Code K Solution
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jee Exam Paper 2 Code K Solution
 - Setting Reading Goals Jee Exam Paper 2 Code K Solution
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jee Exam Paper 2 Code K Solution
 - Fact-Checking eBook Content of Jee Exam Paper 2 Code K Solution
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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