

Stoichiometry Lab: Iron and Copper Sulfate Sample data

Data and observations :

mass of copper (II) sulfate pentahydrate	<u>3.27</u>	g
mass of iron:	<u>0.50</u>	g
mass of filter paper	<u>1.28</u>	g
mass of filter paper and dry copper:	<u>1.84</u>	g
mass of dry copper	<u>0.56</u>	g

Observations of reaction:

When the iron was added a precipitate began to form. The precipitate was reddish in color. The solution remained a blue color but was somewhat lighter in color.

Calculations:

1&2. The single replacement equation will be: $\text{Fe} + \text{CuSO}_4 \longrightarrow \text{FeSO}_4 + \text{Cu}$

3. Fe and CuSO_4 are the reactants and FeSO_4 and Cu are the products.

4. The moles of copper (II) sulfate pentahydrate used were 0.013 mol.

$$3.27 \text{ g CuSO}_4 \cdot 5\text{H}_2\text{O} \times \frac{1 \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}}{249.47 \text{ g CuSO}_4 \cdot 5\text{H}_2\text{O}} = 0.013 \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}$$

5. The moles of iron used were 0.009 mol Fe.

$$0.50 \text{ g Fe} \times \frac{1 \text{ mol Fe}}{55.85 \text{ g Fe}} = 0.009 \text{ mol Fe}$$

6. The moles of copper produced were 0.009 mol Cu.

$$0.56 \text{ g Cu} \times \frac{1 \text{ mol Cu}}{63.55 \text{ g Cu}} = 0.009 \text{ mol Cu}$$

Iron And Copper Sulfate Lab Answers

Michael B. Leyden

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Iron And Copper Sulfate Lab Answers:

The Life Cycle of Everyday Stuff Mike Reeske, Shirley Watt Ireton, National Science Teachers Association, 2001 In the context of life cycles these units use central science concepts to explore the energy raw materials and waste issues that are the history of any manufactured product As students consider the trade offs made at each step they will learn to recognize the decisions made to balance economic developmental and environmental needs **CHEMISTRY EXPERIMENTS** James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction *Laboratory Manual to Accompany Chemistry, [by] Stanley R. Radel, Marjorie H. Navidi* Arthur D. Baker, 1990 *Literature Search* National Library of Medicine (U.S.), 1968

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