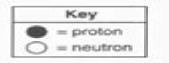
The diagram below represents the nucleus of an atom.





What are the atomic number and mass number of this atom?

- A) The atomic number is 9 and the mass number is 19.
- B) The atomic number is 9 and the mass number is 20.
- C) The atomic number is 11 and the mass number is 19.
- D) The atomic number is 11 and the mass number is 20.
- 2. Which statement concerning elements is true?
 - A) Different elements must have different numbers of isotopes.
 - Different elements must have different numbers of neutrons.
 - C) All atoms of a given element must have the same mass number.
 - D) All atoms of a given element must have the same atomic number.
- 3. Which subatomic particles are located in the nucleus of an He-4 atom?
 - A) electrons and neutrons
 - B) electrons and protons
 - C) neutrons and protons
 - D) neutrons, protons, and electrons
- 4. Which part of a helium atom is positively charged?
 - A) electron

B) neutron

C) nucleus

- D) orbital
- 5. Which total mass is the smallest?
 - A) the mass of 2 electrons
 - B) the mass of 2 neutrons
 - C) the mass of 1 electron plus the mass of 1 proton
 - D) the mass of 1 neutron plus the mass of 1 electron

- Subatomic particles can usually pass undeflected through an atom because the volume of an atom is composed of
 - A) an uncharged nucleus
 - B) largely empty space
 - C) neutrons
 - D) protons
- An experiment in which alpha particles were used to bombard thin sheets of gold foil led to the conclusion that an atom is composed mostly of
 - A) empty space and has a small, negatively charged nucleus
 - B) empty space and has a small, positively charged nucleus
 - C) a large, dense, positively charged nucleus
 - D) a large, dense, negatively charged nucleus
- 8. Which conclusion is based on the "gold foil experiment" and the resulting model of the atom?
 - A) An atom is mainly empty space, and the nucleus has a positive charge.
 - B) An atom is mainly empty space, and the nucleus has a negative charge.
 - C) An atom has hardly any empty space, and the nucleus has a positive charge.
 - D) An atom has hardly any empty space, and the nucleus has a negative charge.
- Which statement about the mass of an electron is correct?
 - A) The mass of an electron is equal to the mass of a proton.
 - B) The mass of an electron is less than the mass of a proton.
 - C) The mass of an electron is equal to the mass of a neutron.
 - D) The mass of an electron is greater than the mass of a neutron.
- 10. Experiments with gold foil indicated that atoms
 - A) usually have a uniform distribution of positive charges
 - B) usually have a uniform distribution of negative charges
 - C) contain a positively charged, dense center
 - D) contain a negatively charged, dense center

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Kathleen Armour

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