Analytical Chemistry MCQ Test. September'21.

- The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
- ammonium hydroxide.

 1. Cu(OH): 2 Zn(OH): 3 Fe(OH): 4 None
- 2. The salt solution which does not react with ammonium hydroxide is ------
- 1. Calcium nitrate 2. Zinc nitrate
- 3.Lead nitrate 4. Copper nitrate
- 3. The metal oxide which can react with acids as well as alkali is------
 - 1. Silver oxide 2. Copper (II) oxide
 - 3. Aluminum oxide 4. Calcium oxide
 - 4.Al metal shows amphoteric behaviour.it can be shown by
 - 1. Its reaction with dil.NaOH only.
 - Its reaction with dil.HCl only.
 - 3. Its reaction with both dil.HCl and dil.NaOH.
 - It does not react with either dil.HCl or dil.NaOH

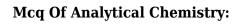
QUESTION 5

Which of the following hydroxides soluble in excess of NaOH?

- 1. Cu (OH)₂ 2. Zn (OH)₂ 3. Fe (OH)₂ 4Fe(OH)₃
- The salt which gives a dirty green precipitate with sodium hydroxide solution is.
 - 1. Iron (III) sulphate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
 - 1. Iron (III) sulphate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
 - 1. Iron (III) choride 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
 - 1. Lead nitrate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate

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Oswald John Simpson



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