

mindray

BeneHeart D6



Monitor OR Power

1 Manual Defib

Energy Select

2 Charge

3 Shock

11 Load Select Alarm Mute CD Load NIBP Event

Large circular button for manual defibrillation/shock delivery.

# Manual Monitor Mindray Beneheart D6

**A Loxley**



## **Manual Monitor Mindray Beneheart D6:**

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$x_a$  is often denoted by  $x_a$  and

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by 100 gives the mole percentage 3 mole fraction is a unitless and dimensionless expression advantages of molar fraction

mole fraction is not dependent on the temperature

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present is equal to one  $x_a + x_b + \dots = 1$  when the mole fractions are multiplied by 100 they give the mole percentage mole fraction is

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mass  $C_6H_6$  6 12 6 1 g mol 1 78 g mol 1 hence  $C_6H_6$  s number of moles 30 78 mol 0 3846 mol molar mass of carbon tet  $CCl_4$  1

12 4 355 154 g mol 1

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of solvent water 100 g 0 1 kg to find molarity of solution solution

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contains 2 92 kg of water 1 molality moles solute divided by kilograms solute 0 449 mol kg x 2 92 kg x 1 31108 mol of kbr 2

moles times molar mass equals grams 1 31108 mol 119 0023 g mol 156 g kbr

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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