

# Journal of Epidemiology & Community Health

## EDITORS' CHOICE

**Socioeconomic differences in acceptability and preferences for policies that aim to reduce socioeconomic inequalities in health**

- Precarious employment and workplace transmission of COVID-19 from worker's compensation claims in Ontario, Canada
- Socioeconomic disparity in the natural history of cutaneous melanoma
- Association of childhood food consumption and dietary pattern with cardiometabolic risk factors and metabolomics in late adolescence

# Journal Of Epidemiology And Community Health Instructions For Authors

**L Darling-Hammond**



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web  $m \times c \times \Delta t$  equation 1 where  $m$  is the total mass of the solution solute plus solvent  $c$  is the specific heat of the solution and  $\Delta t$  is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of *guided inquiry designing the ideal hand warmer* - Dec 27 2022

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution  $\Delta H_{\text{soln}}$  or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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