

$$k = \frac{5}{3}$$

$T_2$  can be calculated by applying the polytropic relation derived for adiabatic expansions.  
From Equation 2.89

$$pV^k = \text{const}$$

$$\therefore p_2 V_2^k = \text{const}$$

The application of the above equation

$$p_1 V_1^k = p_2 V_2^k$$

Since  $T$  is a constant, substitution of the expression for  $T$  into the polytropic relation results in

$$\frac{p_1 V_1^k}{p_1 V_1^k} = \frac{p_2 V_2^k}{p_2 V_2^k}$$

This relation can be used to calculate the  $T_2$

$$T_2 = \text{const} \left( \frac{V_1}{V_2} \right)^{\frac{k-1}{k}}$$

Since that  $T_2$  is a constant, value of  $T_2$  can be calculated

$$T_2 = \text{const} \left( \frac{V_1}{V_2} \right)^{\frac{k-1}{k}}$$

# Koretsky Thermodynamics 2nd Edition Solution Manual

**Y.V.C. Rao**



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