

# Limiting Reactant vs Excess Reactant

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## DEFINITION

### Limiting Reactant

Limiting reactant is the reactant that can limit the formation of the final product in a chemical reaction

### Excess Reactant

Excess reactant is the reactant that is present in excess in a reaction mixture

## DESCRIPTION

Completely consumed during the chemical reaction

Not all reactant molecules are consumed in the reaction

## END OF THE REACTION

No limiting reactant present

Excess reactant is present

## EFFECT ON FINAL PRODUCT

Limits the amount of final product formed

No effect on the final product

# Limiting Reactants Pogil Key

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## **Limiting Reactants Pogil Key:**

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