

Limiting Reactant Practice

5 pages + 5-page Answer KEY + Periodic Table

chemistry

LIMITING REACTANTS

1. The reaction below produces ammonia for use in fertilizers.
 $N_2 + 3H_2 \rightarrow 2NH_3$
 If you start with 10 moles of N_2 and 30 moles of H_2 , determine the limiting reactant.

2. Calculate the grams of ammonia produced.

3. How many grams of the excess reactant is produced as a left over?
 To multiply the moles of the limiting reactant by the mole ratio of the excess reactant.

Calculate moles of excess $\times$ mole of product

4. The reaction below produces carbon dioxide and water from glucose (sugar) and oxygen.
 $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$
 If you have 100 grams of glucose and 60 grams of oxygen, determine the limiting reactant.

5. How many grams of water (steam) are produced?



6. Magnesium chloride is used to create salts for de-icing roads. The reaction with very impure sodium is produced.
 $Mg + 2NaCl \rightarrow MgCl_2 + 2Na$
 If you start with 10 moles of magnesium and 2 moles of sodium, determine the limiting reactant.

7. Calculate the grams of magnesium chloride produced.

8. How many grams of the excess reactant is produced as a left over?

9. The reaction below is used to create the limiting reactant.
 $2H_2 + O_2 \rightarrow 2H_2O$
 If you have 100 grams of hydrogen and 60 grams of oxygen, determine the limiting reactant.

10. How much carbon dioxide is produced?

11. The reaction below produces ammonia for use in fertilizers.
 $N_2 + 3H_2 \rightarrow 2NH_3$
 If you have 100 grams of nitrogen and 30 grams of hydrogen, determine the limiting reactant.

12. Calculate the grams of ammonia produced.
 $N_2 + 3H_2 \rightarrow 2NH_3$
 If you have 100 grams of nitrogen and 30 grams of hydrogen, determine the limiting reactant.

13. How many grams of the excess reactant is produced as a left over?

14. How many moles of carbon dioxide is produced?

15. Calculate the grams of carbon dioxide produced.
 $C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$
 If you have 100 grams of glucose and 60 grams of oxygen, determine the limiting reactant.

16. How many grams of the excess reactant is produced as a left over?

17. Calculate the grams of magnesium chloride produced.
 $Mg + 2NaCl \rightarrow MgCl_2 + 2Na$
 If you have 100 grams of magnesium and 20 grams of sodium, determine the limiting reactant.

18. How many grams of water (steam) are produced?

ANSWER KEY INCLUDED

Limiting Reactant Answer Key

Ralph Thomas Myers



Limiting Reactant Answer Key:

Limiting Reactant Answer Key: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the fascinating narratives that have enthralled audiences this year. Limiting Reactant Answer Key : Colleen Hoover's "It Ends with Us" This touching tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens weaves a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a exceptional and gripping novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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