
Lifeguard Written Test

The primary responsibility of a lifeguard is to ensure _____ and to protect _____. - patron safety and protect lives

In what ways should a professional lifeguard be prepared for his or her job? - Knowledgeable and skilled , reliable, mature, courteous and consistent, positive, professional

What different environments employ lifeguards? - lakes, beaches, coves, pool clubs, community pools

The certificates you earn in this class will prepare you for working in which environment(s)? - *very environment that hires red cross certified lifeguards

List one decision that a lifeguard may have to make? - F.I.N.D

The FIND model stands for: - Figure out the problem, identify the possible solutions, name the pros and cons for each solution, decide which solutions is best

Duty to act ____ - A legal responsibility to respond to an emergency while on the job.

Standard of care - The benchmark by which the actions of a caregiver with similar training and experience is measured.

Negligence ____ - A failure to act or to use reasonable care that results in injury or that causes further harm.

Good Samaritan laws ____ - Statutes that generally provide legal protection to people who willingly (without a duty to act) give emergency care without being negligent or accepting anything in return

Consent ____ - Permission to provide care given by an injured adult or the parent of an injured child to a rescuer or caregiver. This permission is implied when the adult is unresponsive or the child is unattended.

Refusal of care ____ - An unwillingness to receive first aid or medical treatment expressed by a victim (or the victim's parent or guardian) who must give consent.

Abandonment ____ - The failure to continue care once it has been initiated.

Lifeguard Written Test Answers

A Gutmann



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net ionic equation $\text{H}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{Na}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{Na}^+ \text{aq} + \text{Cl}^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g} \rightarrow \text{H}^+ \text{aq} + \text{HCO}_3^- \text{aq} + \text{H}_2\text{O} \text{l} + \text{CO}_2 \text{g}$ here is a video on spectator ions
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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{aq} + \text{OH}^- \text{aq} \rightarrow \text{H}_2\text{O} \text{l}$ the spectator ions in $\text{KOH} \text{aq}$ and $\text{HNO}_3 \text{aq}$ are $\text{K}^+ \text{aq}$ and $\text{NO}_3^- \text{aq}$ how to identify and list the

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