

Manual de Instrucciones

EC 214 - EC 215

Conductivimetro profesional de laboratorio



Entiendo que:

la seguridad que lea el código en pantalla RoRo.

Por favor, lee cuidadosamente este manual antes de utilizar el instrumento. Le proporcionará toda la información necesaria para el correcto uso del mismo. Si necesita información técnica adicional, no dude en contactar con nosotros a través de nuestro e-mail, art@hannainst.com.

Este instrumento cumple con la normativa **CE EN 50601-1** y **EN 50602-1**.

INDICE

| | |
|--|----|
| INSPECCION PRELIMINAR | 3 |
| DESCRIPCION GENERAL | 3 |
| DESCRIPCION FUNCIONAL DEL EC 214 | 4 |
| ESPECIFICACIONES DEL EC 214 | 5 |
| DESCRIPCION FUNCIONAL DEL EC 215 | 6 |
| ESPECIFICACIONES DEL EC 215 | 7 |
| ORDEN DE OPERACIONES | 8 |
| CALIBRACION | 10 |
| CONDUCTIVIDAD Y COEF. DE TEMPERATURA | 12 |
| RENDIMIENTO DE TEMPERATURA | |
| COEFICIENTE DE UNA SOLUCION (COEF. EC 215) | 14 |
| MANTENIMIENTO DE LA SONDA | 15 |
| ACCESORIOS | 16 |
| COMETAS | 18 |
| DETERMINACION DE CONDUCTIVIDAD DE LA CE | 19 |

INSPECCION PRELIMINAR

Extraiga el instrumento de su embalaje original y asegúrese de que no presenta ningún daño debido al transporte. De ser así, consérvelo inmediatamente a su suministrador.

El equipo se suministra con:

- Sonda de conductividad (H06300 para EC 214, H06303 para EC 215)
- Manual de instrucciones;
- Aislador T20C (H070005 o H070006);
- Guía rápida.

Nota: asegure el embalaje hasta que esté completamente seguro de que el instrumento funciona correctamente. Cualquier artículo debido debe ser devuelto en su embalaje original con los accesorios suministrados.

DESCRIPCION GENERAL

EC 214 y EC 215 son conductivímetros profesionales de sobremesa multiusos, diseñados para su uso portátil a la hora de realizar medidas de conductividad en líquidos.

Los 4 rangos de conductividad cubren todo tipo de aplicaciones, desde agua desionizada hasta salinas.

La calibración se realiza de modo sencillo e inmediato incorporando en el modo incorporado en la parte frontal del instrumento.

Las mediciones no precisan la calibración para cualquier rango de medición o otro.

Todas las mediciones en cualquier rango se realizan con la sonda potenciométrica de 4 cables de platina suministrada. Esta sonda es más fácil de limpiar y resistente a la corrosión y temperatura. También sirve para medir líquidos en muestras de pequeño tamaño y se suministra con un cable de 1 m.

El efecto de la temperatura puede ser compensado gracias al modo incorporado en el panel frontal del EC 214.

Con el EC 215, la compensación de la temperatura en los líquidos es automática gracias a la sonda incorporada de 4 cables. El coeficiente de temperatura se regula de 0 a 2.5% gracias al modo incorporado en el panel frontal.

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Este instrumento cumple con los requisitos de la CE

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related laboratory exercise using a similar circuit that is sometimes simplified to make laboratory time as efficient as possible. The same text icon identifies the related DA exercise in the lab manual. One issue is the trend of industry to smaller surface mount devices which are very difficult to work with and are not practical for most lab work. For example, almost all varactors are supplied as surface mount devices now. In reviewing each experiment, we have found components that can illustrate the device function with a traditional one. The traditional through-hole MV2109 varactor is listed as obsolete but will be available for the foreseeable future from Electronix Express www.elexp.com so it is called out in Experiment 3. All components are available from Electronix Express www.elexp.com as a kit of parts; see list in Appendix A. The format for each experiment has not changed from the last edition and is as follows:

- Introduction: A brief discussion about the experiment and comments about each of the independent Parts that follow.
- Reading: Reading assignment in the Floyd text related to the experiment.
- Key Objectives: A statement specific to each Part of the experiment of what the student should be able to do.
- Components Needed: A list of components and small items required for each Part but not including the equipment found at a typical lab station.
- Particular care has been exercised to select materials that are readily available and reusable, keeping cost at a minimum.
- Parts: There are two or three independent parts to each experiment.
- Needed tables, graphs, and figures are positioned close to the first referenced location to avoid confusion.
- Step numbering starts fresh with each Part but figures and tables are numbered sequentially for the entire experiment to avoid multiple figures with the same number.
- Conclusion: At the end of each Part, space is provided for a written conclusion.
- Questions: Each Part includes several questions that require the student to draw upon the laboratory work and check his or her understanding of the concepts.
- Troubleshooting: questions are frequently presented.
- Multisim Simulation: At the end of each experiment except 1, one or more circuits are simulated in a Multisim computer simulation.
- New Multisim troubleshooting problems have been added to this edition.
- Multisim troubleshooting files are identified with the suffix f1, f2, etc. in the file name, standing for fault1, fault2, etc.
- Other files with nf as the suffix include demonstrations or practice using instruments such as the Bode Plotter and the Spectrum Analyzer.
- A special icon is shown with all figures that are related to the Multisim simulation.
- Multisim files are found on the website www.pearsonglobaledition.com.
- Floyd Microsoft PowerPoint slides are available at no cost to instructors for all experiments. The slides reinforce the experiments with troubleshooting questions and a related problem and are available on the instructor's resource site.

Each laboratory station should contain a dual variable regulated power supply, a function generator, a multimeter, and a dual channel oscilloscope. A list of all required materials is given in Appendix A along with information on acquiring the PAM. As mentioned, components are also available as a kit from Electronix Express; the kit number is 32DBEDFL10.

Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16

Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program. The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are

the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don't write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition's experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section

Lab Manual to accompany Physical Science Bill Tillery, 2008-09-05

Lab Manual (Buchla) David M. Buchla, 1998-05-26

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Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 1999-11-17 This flexible lab manual appropriate for use with

a wide range of general chemistry books offers a wealth of practical chemistry experiments It includes pertinent information on rules and safety in the lab Preparation of the new edition was guided by specific feedback from users **Lab Manual to Accompany Physical Science** Bill W. Tillery,Tillery Bill,2006-03 **Lab Manual for General, Organic & Biochemistry** Larry C Byrd, Dr.,Norman W Hunter, Dr.,Charles H Henrickson, Dr.,2010-01-20 The seventh edition by Charles H Henrickson Larry C Byrd and Norman W Hunter of Western Kentucky University offers clear and concise laboratory experiments to reinforce students understanding of concepts Pre laboratory exercises questions and report sheets are coordinated with each experiment to ensure active student involvement and comprehension An updated student tutorial on graphing with Excel has been added to this edition Laboratory Instructor s Manual Written by Charles H Henrickson Larry C Byrd and Norman W Hunter of Western Kentucky University this helpful guide contains hints that the authors have learned over the years to ensure students success in the laboratory This Resource Guide is available through the Connect Chemistry website for this text **Cu in Lab General Chemistry** Dennis L. Stevens,2012-03-05 Fundamentals of Biology Lab Manual Robert Keeton,2021-07-30

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Table of Contents Lab Manual Ec215

1. Understanding the eBook Lab Manual Ec215
 - The Rise of Digital Reading Lab Manual Ec215
 - Advantages of eBooks Over Traditional Books
2. Identifying Lab Manual Ec215
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lab Manual Ec215
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lab Manual Ec215
 - Personalized Recommendations
 - Lab Manual Ec215 User Reviews and Ratings
 - Lab Manual Ec215 and Bestseller Lists
5. Accessing Lab Manual Ec215 Free and Paid eBooks
 - Lab Manual Ec215 Public Domain eBooks

- Lab Manual Ec215 eBook Subscription Services
- Lab Manual Ec215 Budget-Friendly Options
- 6. Navigating Lab Manual Ec215 eBook Formats
 - ePub, PDF, MOBI, and More
 - Lab Manual Ec215 Compatibility with Devices
 - Lab Manual Ec215 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lab Manual Ec215
 - Highlighting and Note-Taking Lab Manual Ec215
 - Interactive Elements Lab Manual Ec215
- 8. Staying Engaged with Lab Manual Ec215
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lab Manual Ec215
- 9. Balancing eBooks and Physical Books Lab Manual Ec215
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lab Manual Ec215
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Lab Manual Ec215
 - Setting Reading Goals Lab Manual Ec215
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lab Manual Ec215
 - Fact-Checking eBook Content of Lab Manual Ec215
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web 2 dna is placed in wells 3 electrical current is ran through gel 4 dna is negative and is attracted to the positive side 5 smallest fragments travel further and banding is compared uses of gel electrophoresis dna relation crime scenes and paternity test which species is most like y to produce curiol z

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