

Experiment 8 Report Sheet

Limiting Reactant

Date _____ Lab Sec. _____ Name _____ Desk No. _____

A. Precipitation of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ from the Salt Mixture

Unknown number _____

	Trial 1	Trial 2
1. Mass of beaker (g)	73.339 g	70.4301 g
2. Mass of beaker and salt mixture (g)	75.1309 g	71.4329 g
3. Mass of salt mixture (g)	1.5762 g	1.0028 g
4. Mass of filter paper (g)	0.3960 g	0.3962 g
5. Mass of filter paper and product after air-dried or oven-dried (g)	0.7609 g	0.70589 g
6. Mass of dried product (g)	0.3647 g	0.30969 g
7. Formula of dried product	CaC_2O_4	CaC_2O_4

B. Determination of Limiting Reactant

1. Limiting reactant in salt mixture (write complete formula)



2. Excess reactant in salt mixture (write complete formula)



Data Analysis

1. Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (or CaC_2O_4) precipitated (mol)	$2.496 \times 10^{-3} \text{ mol}$	$2.120 \times 10^{-3} \text{ mol}$
2. Moles of limiting reactant in salt mixture (mol)	1.980×10^{-3}	1.681×10^{-3}
• formula of limiting hydrate $\text{K}_2\text{C}_2\text{O}_4$		
3. Mass of limiting reactant in salt mixture (g)	0.3647 g	0.30969 g
4. Mass of excess reactant in salt mixture (g)	1.2118 g	0.6931 g
• formula of excess hydrate $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$		
5. Percent limiting reactant in salt mixture (%)	23.12 %	30.88 %
6. Percent excess reactant in salt mixture (%)	76.87 %	69.12 %
7. Mass of excess reactant that reacted (g)	0.2393 g	0.2456 g
8. Mass of excess reactant, unreacted (g)	0.9225 g	0.4475 g

*Show calculations for Trial 1 on next page.

Laboratory Manual Limiting Reactant

Albert A Gayle



Laboratory Manual Limiting Reactant:

This book delves into Laboratory Manual Limiting Reactant. Laboratory Manual Limiting Reactant is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Laboratory Manual Limiting Reactant, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Laboratory Manual Limiting Reactant
- Chapter 2: Essential Elements of Laboratory Manual Limiting Reactant
- Chapter 3: Laboratory Manual Limiting Reactant in Everyday Life
- Chapter 4: Laboratory Manual Limiting Reactant in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Laboratory Manual Limiting Reactant. This chapter will explore what Laboratory Manual Limiting Reactant is, why Laboratory Manual Limiting Reactant is vital, and how to effectively learn about Laboratory Manual Limiting Reactant.

3. In chapter 2, this book will delve into the foundational concepts of Laboratory Manual Limiting Reactant. The second chapter will elucidate the essential principles that need to be understood to grasp Laboratory Manual Limiting Reactant in its entirety.

4. In chapter 3, the author will examine the practical applications of Laboratory Manual Limiting Reactant in daily life. This chapter will showcase real-world examples of how Laboratory Manual Limiting Reactant can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Laboratory Manual Limiting Reactant in specific contexts. The fourth chapter will explore how Laboratory Manual Limiting Reactant is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Laboratory Manual Limiting Reactant. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Laboratory Manual Limiting Reactant.

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