

## Practicing with Isotopes

1. Complete the following questions. Assume all the atoms are neutral.



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_



Element: \_\_\_\_\_  
 # of protons: \_\_\_\_\_  
 # of neutrons: \_\_\_\_\_

2. Calculate the average atomic mass for the following element from the given information about the relevant isotopes:

Isotope 1: Mass = 35 amu  
 Isotope 2: Mass = 37 amu

Abundance = 75.53%  
 Abundance = 24.47%

Which element is this? \_\_\_\_\_

# Isotopes Practice Set Key

**D Kirk**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

## Isotopes Practice Set Key:

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