



Metric Conversion Worksheet Answers

*mm and cm, cm and m, m and km
up to 10 with up to 1 decimal place*

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1) 7000 m = **7** km

2) 1 km = **1000** m

3) 2.4 km = **2400** m

4) **820** cm = 8.2 m

5) 3.1 km = **3100** m

6) 810 cm = **8.1** m

7) 700 cm = **7** m

8) **54** mm = 5.4 cm

9) 7.7 cm = **77** mm

10) **28** mm = 2.8 cm

11) **10** km = 10000 m

12) **3.1** km = 3100 m

13) 4.2 cm = **42** mm

14) **8.8** m = 880 cm

15) **9.5** cm = 95 mm

16) **10300** m = 10.3 km

17) 6.7 km = **6700** m

18) 10.1 m = **1010** cm

19) 7.3 cm = **73** mm

20) **9** m = 900 cm

21) 4.3 m = **430** cm

22) 2 cm = **20** mm

23) **8.7** km = 8700 m

24) 500 cm = **5** m

25) **6.1** m = 610 cm

26) **90** mm = 9 cm

27) **99** mm = 9.9 cm

28) 2 cm = **20** mm

29) **580** cm = 5.8 m

30) **8.2** m = 820 cm

Metric Conversion Practice 2 Answer

Barbara A. Norton, Anna Miller

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author names.

Metric Conversion Practice 2 Answer:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **Simple Steps for Fifth Grade**, 2015-12-14 Simple Steps for Fifth Grade helps your child master math and language arts skills such as multiplication division numbers place value fractions decimals expressions measurement geometry graphing grammar punctuation capitalization usage and sentence structure A standards based resource that simplifies key concepts for easy understanding Simple Steps for Fifth Grade provides learners with easy to follow units clear explanations skill reinforcing activities and an answer key to check accuracy By preparing students for today's rigorous academic standards this comprehensive resource is ideal for supporting classroom learning and enhancing home school curriculum A unique workbook series that offers step by step guidance Simple Steps breaks down essential concepts so that learners can develop a deep understanding of both math and ELA skills for improved academic performance With Simple Steps for Fifth Grade your child is one step closer to complete school success **Pharmacology** Joyce LeFever Kee, Evelyn R. Hayes, Linda E. McCuiston, 2014-01-30 Previous edition has subtitle a nursing process approach **Pharmacology - E-Book** Linda E. McCuiston, Joyce LeFever Kee, Evelyn R. Hayes, 2014-02-03 NEW QSEN focus emphasizes patient centered care safety quality and collaboration and teamwork NEW content covers the most commonly used drugs including updated Prototype Drug Charts and drug tables Enhanced coverage of prioritization includes nursing interventions in the Nursing Process sections listed in order of priority Updated illustrations include new drug labels in the Drug Calculations chapter **Investigating Chemistry** Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly

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