

CHAPTER 2 REVIEW

Measurements and Calculations

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model.

_____ observation and data

a. A research team records the rainfall in inches per day in a prescribed area of the rain forest. The square footage of vegetation and relative plant density per square foot are also measured.

_____ observation and data

b. The intensity, duration, and time of day of the precipitation are noted for each precipitation episode. The types of vegetation in the area are recorded and classified.

_____ control

c. The information gathered is compared with the data on the average precipitation and the plant population collected over the last 10 years.

_____ hypothesis

d. The information gathered by the research team indicates that rainfall has decreased significantly. They propose that deforestation is the primary cause of this phenomenon.

2. "When 1.00 g of a white, crystalline sugar are dissolved in 100 mL of water, the solution is observed to freeze at -0.54°C , not 0.0°C . The system is denser than pure water." Which parts of these statements represent quantitative information, and which parts represent qualitative information?

Quantitative values include the mass of sugar, volume of water, and observed
freezing point. Qualitative properties are the color and state of the sugar and the
claim of greater density.

3. Compare and contrast a model with a theory.

Theories are broad generalizations used to explain observations. Models are
a physical object used to illustrate or explain complex concepts or an explanation of
how phenomena occur and how data and events are related.

Measurements And Calculations Concept Review Answer Key

Lei Shi



Measurements And Calculations Concept Review Answer Key:

Using Basic Statistics in the Behavioral and Social Sciences Annabel Ness Evans, 2013-06-06 Using Basic Statistics in the Behavioral and Social Sciences Fifth Edition by Annabel Ness Evans presents introductory statistics in a practical conceptual and humorous way reducing the anxiety that many students experience in introductory courses Avoiding complex notation and derivation the book focuses on helping readers develop an understanding of the underlying logic of statistics Practical Focus on Research boxes engage students with realistic applications of statistics and end of chapter exercises ensure student comprehension This exciting new edition includes a greater number of realistic and engaging global examples within the social and behavioral sciences making it ideal for use within many departments or in interdisciplinary settings Holt Chemistry Holt Rinehart & Winston, 2003-01-24 *Chemistry3* Andrew Burrows, John Holman, Simon Lancaster, Andrew Parsons, Tina Overton, Gwen Pilling, Gareth Price, 2021 Chemistry is widely considered to be the central science it encompasses concepts on which all other branches of science are developed Yet for many students entering university gaining a firm grounding in chemistry is a real challenge Chemistry3 responds to this challenge providing students with a full understanding of the fundamental principles of chemistry on which to build later studies Uniquely amongst the introductory chemistry texts currently available Chemistry3's author team brings together experts in each of organic inorganic and physical chemistry with specialists in chemistry education to provide balanced coverage of the fundamentals of chemistry in a way that students both enjoy and understand The result is a text that builds on what students know already from school and tackles their misunderstandings and misconceptions thereby providing a seamless transition from school to undergraduate study Written with unrivalled clarity students are encouraged to engage with the text and appreciate the central role that chemistry plays in our lives through the unique use of real world context and photographs Chemistry3 tackles head on two issues pervading chemistry education students mathematical skills and their ability to see the subject as a single unified discipline Instead of avoiding the maths Chemistry3 provides structured support in the form of careful explanations reminders of key mathematical concepts step by step calculations in worked examples and a Maths Toolkit to help students get to grips with the essential mathematical element of chemistry Frequent cross references highlight the connections between each strand of chemistry and explain the relationship between the topics so students can develop an understanding of the subject as a whole Digital formats and resources Chemistry3 is available for students and institutions to purchase in a variety of formats and is supported by online resources The e book offers a mobile experience and convenient access along with functionality tools navigation features and links that offer extra learning support www.oxfordtextbooks.co.uk/ebooks The e book also features interactive animations of molecular structures screencasts in which authors talk step by step through selected examples and key reaction mechanisms and self assessment activities for each chapter The accompanying online resources will also include for students DT Chapter 1 as an open access PDF DT Chapter summaries and key equations to

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