

## JUNE 2010 – THE START OF THE NEW STYLE PAPER

### For 20 mark questions

Level	Mark	Descriptor
Level 1	1-7	Identification of points which are largely irrelevant
Level 2	8-10	Typically, answers will consider 1 factor with 1 evaluative point or 2 points with no evaluation or list of several points with little analysis
Level 3	11-14	Typically, answers will consider 2 factors with 1 evaluative point or 3 points with no evaluation
Level 4	15-16	Typically, answers will consider 3 factors with 1 evaluative point
Level 5	17-20	Typically, answers will consider 3 factors with at least 2 evaluative points

### GUIDELINES FOR MARKING ESSAY QUESTIONS 6EC04 June 2012

Level	Part (a) Questions: Performance Criteria for Mark base 20	
Level 1	1-7	<ul style="list-style-type: none"> <li>Displays knowledge presented as facts without awareness of other viewpoints</li> <li>Demonstrates limited understanding with little or no analysis</li> <li>Attempts at selecting, applying, evaluating and presenting material which are irrelevant and unclear</li> <li>None of the assessment objectives are covered satisfactorily</li> </ul> <p>Identification of points which are largely irrelevant</p>
Level 2	8-10	<ul style="list-style-type: none"> <li>Displays elementary knowledge of well learnt economic facts showing a generalised understanding together with limited analysis i.e. identification of points or a very limited discussion</li> <li>Displays a limited ability to select and apply different economic ideas</li> <li>Evidence presented has a basic relevance</li> </ul> <p>Typically, answers will consider 1 factor with 1 evaluative point or 2 points with no evaluation or list of several points with little analysis</p>
Level 3	11-14	<ul style="list-style-type: none"> <li>Displays knowledge and understanding of economic principles, concepts and theories as well as some analysis of issues i.e. answer might lack sufficient breadth and depth to be worthy of a higher mark</li> <li>Shows some ability to apply economic ideas and relate them to economic problems</li> <li>Employs different approaches to reach conclusions, presenting evidence with some relevance and coherence</li> </ul> <p>Typically, answers will consider 2 factors with 1 evaluative point or 3 points with no evaluation</p>
	11-12	Award in this range for 2 factors and 1 evaluative comment (or 3 factors with no evaluation) which are not well developed/reasoned
	13-14	Award in this range for 2 factors and 1 evaluative comment (or 3 factors with no evaluation) which are well developed and reasoned
Level 4	15-16	<ul style="list-style-type: none"> <li>Displays a good knowledge of economic principles, concepts and theories together with an analysis of the issues involved</li> <li>Demonstrates an ability to select and apply economic ideas and to relate them to economic problems</li> <li>Evidence of some evaluation of alternative approaches leading to conclusions which are presented in a relevant and coherent way</li> </ul> <p>Typically, answers will consider 3 factors with 1 evaluative point</p>
Level 5	17-20	<ul style="list-style-type: none"> <li>Displays a wide range of knowledge of economic principles, concepts and theories together with a rigorous analysis of issues</li> <li>Demonstrates an outstanding ability to select and apply economic ideas to economic problems</li> <li>Evaluation is well balanced and critical leading to valid conclusions</li> <li>Material is presented in a relevant, clear and coherent way with evidence fully and reliably integrated</li> <li>Excellence is displayed across all assessment objectives</li> </ul> <p>Typically, answers will consider 3 factors with 2 evaluative points</p>

# June 2013 M2 Mark Scheme

**Assessment and Qualifications  
Alliance**



## June 2013 M2 Mark Scheme:

Research Handbook on Climate Change Mitigation Law Geert Van Calster, Wim Vandenberghe, Leonie Reins, 2015-01-30 As well as taking stock of the current and proposed legal instruments the book looks at the wider policy and economic aspects of coping with climate change It provides a comparative overview of key issues across Europe the United States Asia Pacific

**IGCSE Mer Merlyn**, 2015-08-27 Buku IGCSE ini merupakan aplikasi dari pelajaran matematika yang berbasis di University of Cambridge Berisi kumpulan soal lengkap dengan kunci jawaban Memudahkan para siswa dalam menghadapi pelajaran matematika Semoga buku ini bisa menjadi jembatan bagi para siswa yang ingin menguasai mata pelajaran matematika dengan baik

**Energy Law in Taiwan** Anton Ming-Zhi Gao, 2021-05-20 Derived from the renowned multi volume International Encyclopaedia of Laws this book provides a systematic approach to legislation and legal practice concerning energy resources and production in Taiwan The book describes the administrative organization regulatory framework and relevant case law pertaining to the development application and use of such forms of energy as electricity gas petroleum and coal with attention as needed to the pervasive legal effects of competition law environmental law and tax law A general introduction covers the geography of energy resources sources and basic principles of energy law and the relevant governmental institutions Then follows a detailed description of specific legislation and regulation affecting such factors as documentation undertakings facilities storage pricing procurement and sales transportation transmission distribution and supply of each form of energy Case law intergovernmental cooperation agreements and interactions with environmental tax and competition law are explained Its succinct yet scholarly nature as well as the practical quality of the information it provides make this book a valuable resource for energy sector policymakers and energy firm counsel handling cases affecting Taiwan It will also be welcomed by researchers and academics for its contribution to the study of a complex field that today stands at the foreground of comparative law

Parliamentary Debates (Hansard). Great Britain. Parliament. House of Commons, 2013

Advances and Trends in Engineering Sciences and Technologies II Mohamad Ali, Peter Platko, 2016-11-30 These are the proceedings of the 2nd International Conference on Engineering Sciences and Technologies ESaT 2016 held from 29th of June until the 1st of July 2016 in the scenic High Tatras Mountains Tatransk Matliare Slovak Republic After the successful implementation and excellent feedback of the first international conference ESaT 2015 ESaT 2016 was organized under the auspices of the Faculty of Civil Engineering Technical University of Košice Slovak Republic in collaboration with the University of Miskolc Hungary The conference focused on a wide spectrum of topics and subject areas in civil engineering sciences The proceedings bringing new and original advances and trends in various fields of engineering sciences and technologies that accost a wide range of academics scientists researchers and professionals from universities and practice The authors of the articles originate from different countries around the world guaranteeing the importance topicality quality and level of presented results

*Mountain Building* György Hetényi, Mark R. Handy, A. Alexander G.

Webb,2021-12-22      **Wildlife Conservation on Farmland Volume 1** David W. Macdonald,Ruth E. Feber,2015-07-30  
 Using more than 30 years research from the author team at the Wildlife Conservation Research Unit WildCRU this volume reveals how agricultural systems and wildlife interact presenting examples from scales varying from landscape to microcosm from populations to individuals covering plants invertebrates birds and mammals It demonstrates the essential ecosystem services provided by agricultural land and discusses the implications of agricultural development for natural habitats and biodiversity      Travelor's Guide to Ship Museums in the United States Charles L. Barnett,2025-09-12 This book comes from the author s passion for traveling learning and exploring new adventures Additionally a love ships of many types from the US Navy the US Coast Guard other nations and the contribution of all of them to the history of the United States of America This book s intention is to provide the reader with a guide to plan trips to the ship museums in the United States Although there seems to be a rather small market for coffee table books this book is intended as such The first part is organized to list the ship museums as follows 1 by state 2 by city and 3 by ship s name The second part contains the appendixes which organize the ships to aid the reader as follows Database One listing is as follows 1 by state 2 by city 3 by hull number if it applies 4 by page number 5 by telephone number 6 by email address 7 by ship type and 8 by information source Database Two listing is as follows 1 ships name in more than one way for example USS Alabama and Alabama USS 2 by hull number if it applies 3 by state 4 by city 5 by page number 6 by phone number and 7 by email address The sources of the information are listed in the preface of this book      Solar Technologies for the 21st Century Anco S. Blazej,2021-01-07 This book examines solar technologies describes their properties and evaluates the technological potential of each It also reviews the logistics of deploying solar energy as a viable and sustainable way to solve urgent energy environmental and socio economic problems Topics discussed include solar power generation today s solar technologies solar thermal silicon PV thin PV 3 D solar cells nano PV organic solar cells solar successes and failures solar power fields finance and regulations solar markets and solar energy and the environment      The Chemical Engineer ,2008      Water and Energy International ,2016      Pakistan Economic Survey ,2012      Planning ,2008-04      Mark Scheme, Mathematics A (5301, 5306, 5311, 6301,& 6311) Assessment and Qualifications Alliance,2003      13+ Mathematics Mark Scheme (Summer Term 2009) Iseb Iseb,2009-06-08      **Mathematics Tests** ,2012      **13+ Mathematics Mark Scheme (Spring Term 2009)** Iseb Iseb,2009-02-08      1993 Key Stage 3 Mathematics Tests Mark Scheme School Examinations and Assessment Council (Great Britain),1993

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web isotopes differ in the number of neutrons in ions the number of electrons is different from the number of protons

isotopes are atoms that have the same number of protons but different numbers of neutrons thus atoms of  $^{12}_6\text{C}$  and of  $^{13}_6\text{C}$  are isotopes of each other

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web what is the percent composition of Cu in terms of these two isotopes answer 69.15% Cu-63 and 30.85% Cu-65 figure

pageindex 5 isotopes of an element are atoms with the same atomic number but different mass numbers isotopes of an

element therefore differ from each other only in the number of neutrons within the nucleus

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web jan 11 2013 isotopes of an element share the same number of protons but have different numbers of neutrons let's use carbon as an example there are three isotopes of carbon found in nature carbon 12

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web video answer so now we'll talk about problems 58 from chapter four this problem whereas how do isotopes differ and how are they similar isotopes for the same element so isotopes differ in the number of neutrons

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web isotopes are atoms of the same element that differ in the amount of neutrons and atomic mass almost all elements on the periodic table have at least two different natural isotopes many elements

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web the isotopes of a given element always contain the same number of protons and therefore occupy the same place on the periodic table however because isotopes contain different numbers of neutrons each isotope has a unique atomic mass

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web apr 18 2015 isotopes are variants of a particular chemical element which differ in neutron number although all isotopes of a given element have the same number of protons in each atom answer link

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web jul 25 2014 isotopes differ in the number of neutrons in ions the number of electrons is different from the number of protons isotopes are atoms that have the same number of protons but different numbers of neutrons thus atoms of  $^{12}_6\text{C}$  and  $^{13}_6\text{C}$  are isotopes of each other

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web isotopes are those atoms having same atomic number number of protons are same but different mass number number of neutrons differ ions are atoms which contain an overall charge where number of protons number of electrons

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web atoms of the same element i.e. same atomic number  $Z$  that have different numbers of neutrons are called isotopes for example 99 of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web aug 9 2023 how do different isotopes of an element differ two isotopes differ in the number of neutrons in the nucleus for example carbon 14 has 8 neutrons differing from carbon 12 which has 6

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web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number ie same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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web may 16 2018 explanation an atom is characterized as a particular element by the number of positively charged massive nuclear particles i e  $z$   $z=1$  we got hydrogen  $z=2$  we got helium  $z=3$  lithium  $z=4$  we got silver the nucleus can also contain neutrons massive nuclear particles of zero charge

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web atoms of the same element with different numbers of neutrons are called isotopes close isotope atoms of an element with the same number of protons and electrons but different numbers of neutrons

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties every chemical element has one or more isotopes

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus

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web describe the training effect and how it affects the lungs during exercise training effect predicts how your efforts will impact your aerobic performance capacity by exercising the muscles surrounding your lungs increases their strength and

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