

### 3. Marking points

#### 3.1 Marking of lists

This applies to questions requiring a set number of responses, but for which students have provided extra responses. The general principle to be followed in such a situation is that 'right + wrong = wrong'.

Each error / contradiction negates each correct response. So, if the number of error / contradictions equals or exceeds the number of marks available for the question, no marks can be awarded.

However, responses considered to be neutral (indicated as \* in example 1) are not penalised.

Example 1: What is the pH of an acidic solution?

[1 mark]

Student	Response	Marks awarded
1	green, 5	0
2	red*, 5	1
3	red*, 8	0

Example 2: Name two planets in the solar system.

[2 marks]

Student	Response	Marks awarded
1	Mercury, Venus, Earth, Mars, Jupiter, Saturn, Uranus, Neptune	0

#### 3.2 Marking of extended responses (questions)

For questions requiring an extended response, the whole response of a candidate is marked. The whole of the given answer is marked and not just the correct parts of the answer. Each error is penalised.

#### 3.3 Marking procedures for calculations

Errors should be marked for each stage of the calculation completed correctly. An incorrectly calculated answer does not mean that the whole calculation is marked as given for a correct answer. Errors without any working attract 0 marks.

#### 3.4 Interpretation of 0

Answers using the word 0 should be given credit only if it is clear that the 0 refers to the correct value.

# Jan 2014 Chemistry 2 Mark Scheme

**R Barnett**



**Jan 2014 Chemistry 2 Mark Scheme:**

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