

# Student Example Lab Report

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## Title:

Intermolecular Strengths of Substances

## Objective(s):

- To compare the boiling points of three substances
- To infer relative intermolecular strengths of substances using their boiling points

## Hypothesis:

I predict that vinegar has stronger intermolecular forces, and therefore a higher boiling point, compared to apple juice.

## Procedure:

### Materials: (List all the items used)

- Two safe substances to boil (apple juice, orange juice, milk, vinegar, cooking wine, saltwater)
- Heat source (stove or hot plate)
- Measuring cup
- Thermometer
- Container for heating (pot or beaker)

### Variables: (List the controlled, independent, and dependent variables of the lab.)

1. The controlled variables are the heat source, thermometer, container type, volume of liquid, and the boiling point of water (100 °C or 212 °F).
2. The independent variable (the variable we are changing) is the type of liquid heated.
3. The dependent variable (the variable that changes due to the independent variable) is the boiling point of each liquid.

### Summary of Steps:

1. Select two safe substances to boil. Make sure to list the units used to measure your substance when recording your volumes in the data chart.
2. Measure ½ to a full cup of your first substance and pour it into your heating container.

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**Robert R. Redfield**



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