1. (4 points)

The specific heat (s) of iron (Fe) is 0.40 J/(g K). Calculate the heat effect q when 20 g of iron is heated from 100 °C to 200 °C. Include the sign!

 $q = m.s.\Delta T$ for $\Delta T \Delta K = \Delta^{\circ}C$ q = 20x0.40x(200-100) = +800 J

You must include the units and sign in the answer!

2. (4 points)

Reaction: $Mg(s) + \frac{1}{2}O_3(g) \rightarrow MgO(s)$ $\Delta H = -600 \text{ kJ/mol}$

The Molar Mass of Mg = 24.0 g/mol.

Calculate the enthalpy change when 2.40 g Mg(s) reacts with excess oxygen. Include the sign!

 $2.40 \text{ g Mg} = 2.40/24.0 = 0.1 \text{ mol} \rightarrow \Delta H = 0.1(\text{mol})x(-600)(kJ/\text{mol}) = -60 \text{ kJ}$

Again, your answer must include both the sign and the units.

 (2 points) Which one of the following four reactions is a formation reaction? <u>Circle</u> the correct answer.

a)
$$H_2O(I) + \frac{1}{2}O_2 \longrightarrow H_2O_2(I)$$

b)
$$SO_2(g) + \frac{1}{2}O_2 \rightarrow SO_2(g)$$

c)
$$Cl(g) + H(g) \rightarrow HCl(g)$$

d)
$$S(s) + O_2(g) \rightarrow SO_2(g)$$

Look up the definition of a formation reaction in Chang or in your lecture notes: the compound is formed from its elements, and the elements must be as they naturally occur, so e.g. Cl_2 not Cl_4 and H_2 not H_3 !

CHEM 101 Section L52 Quiz #6 b

1. (4 points)

The specific heat (s) of copper (Cu) is 0.38 J/(g.K). Calculate the heat effect q when 10 g of copper is heated from 100 °C to 300 °C. Include the sign!

 $q = m.s.\Delta T$ for $\Delta T \Delta K = \Delta^*C$ q = 10x0.38x(300-100) = +760 J

You must include the units and sign in the answer!

2. (4 points)

Reaction: $Ca(s) + \frac{1}{2}O_2(g) \rightarrow CaO(s)$

$$\Delta H = -650 \text{ kJ/mol}$$

The Molar Mass of Ca = 40.0 g/mol.

Calculate the enthalpy change ΔH when 8.0 g Ca(s) reacts with excess oxygen. Include the sign!!

 $8.0 \text{ g Ca} = 8.0/40/ = 0.2 \text{ mol} \rightarrow \Delta H = 0.2(\text{mol})x(-650)(kJ/\text{mol}) = -130 \text{ kJ}$

Again, your answer must include both the sign and the units.

 (2 points) Which one of the following reactions is a formation reaction? <u>Circle</u> the correct answer.

a)
$$Br(g) + H(g) \rightarrow HBr(g)$$

b)
$$H_2O(I) + \frac{1}{2}O_2 \rightarrow H_2O_2(I)$$

c)
$$2P(s) + 2\%O_2(g) \rightarrow P_2O_3(s)$$

d)
$$SO_2(g) + \frac{1}{2}O_2 \rightarrow SO_3(g)$$

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