

Chemistry May 2012 Paper 2A no.12a) Sodium Chloride (NaCl):

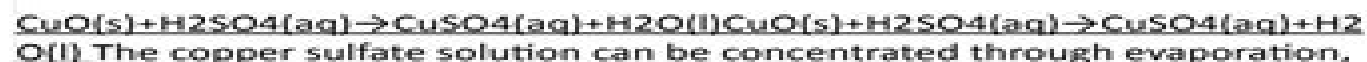
Sodium chloride can be prepared by the reaction of hydrochloric acid (HCl) with sodium hydroxide (NaOH) in a neutralization reaction. The balanced chemical equation is:



To obtain solid sodium chloride, the resulting solution can be evaporated to dryness, leaving behind the salt.

b) Copper (II) Sulfate (CuSO<sub>4</sub>):

Copper (II) sulfate can be obtained by reacting copper oxide (CuO) with sulfuric acid (H<sub>2</sub>SO<sub>4</sub>). The balanced chemical equation is:



The copper sulfate solution can be concentrated through evaporation, and upon cooling, solid crystals of copper (II) sulfate will form. Filtration is then used to separate the crystals from the remaining solution.

c) Potassium Nitrate (KNO<sub>3</sub>):

Potassium nitrate can be prepared by the reaction of nitric acid (HNO<sub>3</sub>) with potassium hydroxide (KOH). The balanced chemical equation is:

d) Calcium Carbonate (CaCO<sub>3</sub>):

Calcium carbonate can be prepared by mixing solutions of calcium chloride (CaCl<sub>2</sub>) and sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>). The balanced chemical equation is:

# Mc Chemistry May 2012 Paper 1 6ch01

**Katrin Zwirgmaier**



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