

## THE SPECTROSCOPIC IDENTIFICATION OF ORGANIC FUNCTIONAL GROUPS

### IR and NMR Spectroscopy

1. The following is part of an IR spectrum (approximate reading):
- a) Absorption bands (approximate values):
  - b) Assigning the compounds to name the compound(s) identified.
  - c) A molecular formula.
  - d) The name of the compound.
2. A compound with molecular formula  $C_8H_{10}O$  has IR spectrum (approximate reading) as follows:
- a) Strongest bands of absorption (approximate):
  - b) Strongest bands of absorption (approximate):
  - c) The intensity of peaks.
  - d) The peak positions (cm<sup>-1</sup>).
3. Which one of the following IR bands is equivalent to the band in the IR spectrum?
- a) 1715 cm<sup>-1</sup>
  - b) 1735 cm<sup>-1</sup>
  - c) 1755 cm<sup>-1</sup>
  - d) 1775 cm<sup>-1</sup>
4. The IR spectrum of a compound shows a strong absorption band at 1715 cm<sup>-1</sup>.
- a) The compound is an aldehyde.
  - b) The compound is a ketone.
  - c) The compound is an alcohol.
  - d) The compound is an amine.
5. The IR spectrum of a compound shows a strong absorption band at 1715 cm<sup>-1</sup>.
- a) The compound is an aldehyde.
  - b) The compound is a ketone.
  - c) The compound is an alcohol.
  - d) The compound is an amine.

# Kmlttb May 2015 Exms

**Roman Wölfel**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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