

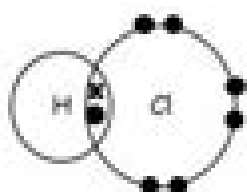
C3 - STRUCTURE AND BONDING 3.5 Covalent bond

Aiming for Grade 4

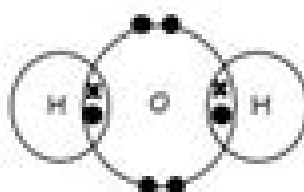
Ionic bonds form between **metals** and **non-metals**. Covalent bonds form between **non-metals**. Covalent bonds form when atoms **share** pairs of **electrons**. These bonds are very **strong**.

Share, Metals, strong, non-metals, electrons
You can use the words more than once.

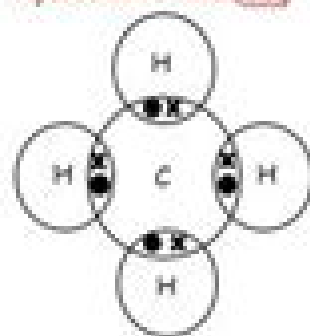
Name these small molecules and state their formulae.



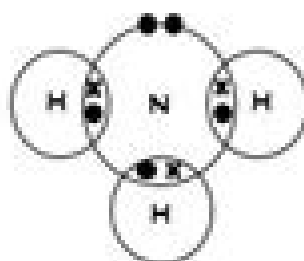
Hydrochloric Acid (HCl)



Water (H_2O)



Methane (CH_4)



Ammonia (NH_3)

Assessment

Aiming for Grade 6

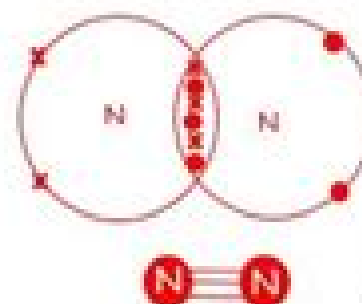
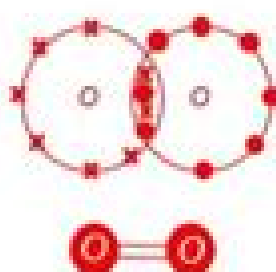
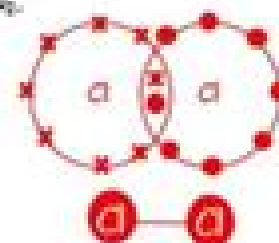
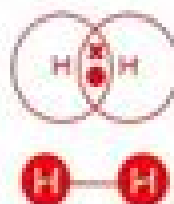
Describe how Hydrogen and Chlorine form a covalent bond to form Hydrochloric acid.

To gain full outer shells Hydrogen needs to gain an electron and Chlorine needs to lose one. Hydrogen shares its outer shell electron with Chlorine and Chlorine shares one of its electrons.

What is a double covalent bond?

Two shared pairs of electrons

Draw dot and cross diagrams and ball and stick diagrams for H_2 , Cl_2 , O_2 , and N_2 .



Assessment

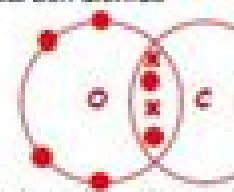
Aiming

Draw ball and stick or the following covalent

Three hydrogen atoms
one phosphorus (phos)



Carbon dioxide



Ethane (2 carbon with hydrogen)



How do double and triple properties?

A double or triple bond the bond making the me

Assessment
Grade 6

1. I can draw dot and

June 2013 Aqa Chemistry C3 Mark Scheme

Richard Bailey



June 2013 Aqa Chemistry C3 Mark Scheme:

Reviewing **June 2013 Aqa Chemistry C3 Mark Scheme**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**June 2013 Aqa Chemistry C3 Mark Scheme**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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