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### Passage 1 (Questions 1-4)

To test how amino acids first appeared on this planet, an experiment was set up to simulate the conditions thought to have existed on early Earth.  $\text{H}_2\text{O}$ ,  $\text{CH}_4$ ,  $\text{NH}_3$ , and  $\text{H}_2$  were all introduced into a sterile arrangement of sealed flasks and tubes. Liquid water evaporated in one compartment as “lightning” was induced between a pair of electrodes. The water vapor then cooled and condensed in a connected section, which fed back to the first compartment, until a dynamic equilibrium was reached.

After two weeks of continuous operation, roughly 2% of the carbon in the system was in the form of amino acids, with glycine being the most abundant. The following steps are believed to account for the formation of the key intermediates, formaldehyde and hydrogen cyanide.



#### Mechanism 1

Via the Strecker Synthesis, glycine can then form as follows:



#### Mechanism 2

Several other amino acids were also observed to form. Of the chiral amino acids, both enantiomers were created in a racemic mixture, though in most biological systems L-amino acids are used exclusively for protein synthesis.

Simple sugars also formed in the same environment. Once formaldehyde is present to react with  $\text{H}_2\text{O}$ , it does so according to the Butlerov reaction:



Figure 1. Formation of carbohydrates from aqueous formaldehyde

### Question 4

Which of the following will increase the yield of carboxylic acid formed from a nitrile group-containing compound, using a mechanism similar to the Strecker synthesis?

- ☐ A. Use a stronger nucleophile, like a Grignard Reagent, instead of  $\text{H}_2\text{O}$
- ☐ B. Attach an electron withdrawing group to the carbon of the nitrile group
- ☐ C. Use a stronger electrophile, like a Grignard Reagent, instead of  $\text{H}_2\text{O}$
- ☐ D. Attach an electron donating group to the carbon of the nitrile group

# Kaplan Mcat Practice Test Answers

**Mark Steyvers**



## **Kaplan Mcat Practice Test Answers:**

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The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts.

The Secret History is a masterful and gripping novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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