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CHEMISTRY PAPER-II

(MODEL PAPER)

Annual Examination 2021

Total Time: 2 hours.

Total Marks: 85

Time :30 min **SECTION 'A' (M.C.Qs. (Multiple Choice Questions))** Marks: 43

Note: This section consists of 43 questions. Attempt all M.C.Qs. Each carries 1 marks.

Q 1:- Choose the correct answers for each from the given options:

The element which is present in group V A and period 3rd, its atomic number is:

- * 15 * 7 * 8 * 18

2) Hydrides of group V A are ----- in nature:

- * Acidic * Basic * Amphoteric * Neutral

3) Potassium when combines with oxygen form:

- * Normal oxide * Super oxide * Per oxide * All

4) Lithium has many similarities to its diagonal neighbour-----in "Be" family:

- * K * Mg * Ca * Na

5) $\text{Li}^+_{\text{aq}}/\text{Li}$ couple has exceptionally high negative electrode potential because of its large value of

- * Ionization potential * Hydration enthalpy
* Electron Affinity * Electronegativity

6) Aluminium does not react with Nitric acid due to

- * Low reactivity * High Ionization Potential
* Formation of oxide layer * It is a metal

7) "A" is the element of III A group which "B" belongs to V A group. When "A" reacts with "B" forms:

- * A_2B_3 * AB * AB_3 * A_3B_2

8) The mixture of "Al" and Fe_2O_3 is used in :

- * Pyrolysis * Thermite process * Electrolysis * Washing

9) This element is solid at room temperature and pressure:

- * Oxygen * Fluorine * Bromine * Iodine

10) Electronic configuration of Cu^{+1} :

- * $4s^1, 3d^{10}$ * $4s^0, 3d^9$ * $4s^0, 3d^{10}$ * $4s^2, 3d^{10}$

11) The coordination number of Pt in $[\text{Pt}(\text{en})_2\text{Cl}_2]^{+2}$ is:

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