## **Acids and Bases**

Question 1:		Choose the correct answer for each of the following questions:				
1.1	The element present in all acids					
	A	oxygen	В	carbon o	dioxide	
	C	nitrogen	D	hydre		
1.2	A sample of an acid in a beaker has a pH of 4. An alka.					pH of
	the liquid in the beaker?					
	A	decreases	В	stays the	e same	
	C	Increases	D	neutral		
1.3	Choose the incorrect combination:					
	A	HNH₄OH – Nitric acid.	В	H <sub>2</sub> SO	sulphuric ac	
	C	HCl - Hydrochloric acid.	D		rbonic acid.	
1.4	The colour of blue litmus paper in an acid is					
	A	yellow	В	en		
	C	blue				
1.5	Which substance is always produced in eaction. ochloric and and lithium?					
	A	water		hydrogen 8-		
	C	oxygen gas		on dioxide gas		
1.6	Which:	substance is always produ in	the rea.		n hydrochloric acid and sodium	
	hydroxi	ide?				
	A	water		hydrogen gas		
	C	oxygen gas		ecipitate		
1.7	Which formula repr					
	A	кон		KCI		
	C	СН₃ОН		CH₃COOH		
1.8	T	od of adding an a	irop b	y drop:		
		neutralisation	В	precipita	ation	
		eva-	D	decomp	osition	[8]
Quest.	.ne genera reaction for an			acid with	the following compounds:	
2.1	Base:					
2.2	Carbonate:					
2.3	Metal:			-(8)-5	turdvehamo-co.	za

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John Conrad Almack, Albert Ray Lang

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