

Answers to the Student book

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1 States of matter

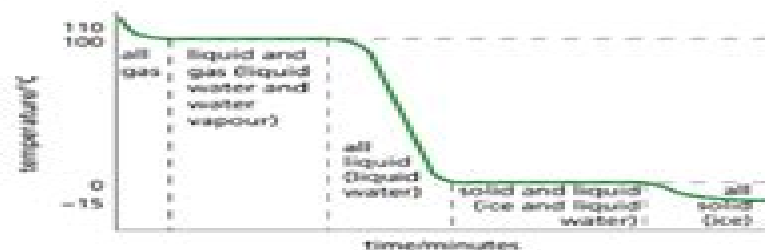
Test Yourself Questions

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- 1 The particles in copper are vibrating about a fixed point. As the temperature increases, they gain kinetic energy and vibrate more, colliding with neighbouring particles. This has the effect of increasing the overall volume of the metal. In other words, it expands.

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- 2 The particles in gases have much weaker forces of attraction between their particles as the particles are already further away from each other compared with those in a solid. These weaker forces of attraction are more easily broken as temperatures rise, allowing the particles to move away from one another more easily.
- 3 There is friction between the ice and the air as the car moves through it. This friction produces heat which causes the ice to melt.
- 4 Adding salt to ice lowers the freezing point of the ice. As long as the temperature is not lower than the freezing point of the mixture of salt and ice, the ice will melt.
- 5



The graph should show that when the changes of state occur the temperature does not change until the change of state is complete. So there should be a flat line at 100°C as the steam condenses to form water and at 0°C as the water freezes to form ice.

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