

# 13.1

## THE NATURE OF GASES

### Section Review

#### Objectives

1. describe the characteristics of ideal gases and real gases
2. distinguish between an ideal gas and a real gas
3. define the macroscopic terms appropriate to gaseous states

#### Vocabulary

- |              |                |                  |
|--------------|----------------|------------------|
| a. ideal gas | b. pressure    | c. pressure      |
| d. real gas  | e. temperature | f. volume        |
| g. gas       | h. density     | i. ideal gas law |
| j. gas       | k. density     | l. ideal gas law |

#### Part A Completion

Use the vocabulary terms in parentheses to complete each of the sentences. Write the complete sentence on the lines provided.

1. Gases have particles that are \_\_\_\_\_, moving constantly in random motion. \_\_\_\_\_ the gas expands to \_\_\_\_\_ the \_\_\_\_\_ of the container. \_\_\_\_\_ the gas particles are \_\_\_\_\_, the \_\_\_\_\_ of the gas is \_\_\_\_\_, and the \_\_\_\_\_ of the gas is \_\_\_\_\_.

1
2
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#### Part B True-False

Write **T** for true or **F** for false on the lines provided.

1. Gases have particles that are \_\_\_\_\_.
2. Gases have particles that are \_\_\_\_\_.

# Nature Of Gases Section Review Answer Key

**David Baud**



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