

## 13.1

## THE NATURE OF GASES

## Section Review

## Objectives

1. describe the characteristics of matter having gaseous properties
2. distinguish between a gas and vapour
3. define the macroscopic and kinetic approaches to understanding gases

## Vocabulary

- |                 |                         |                  |
|-----------------|-------------------------|------------------|
| a. condensing   | b. vacuum               | c. pressure      |
| d. condensation | e. atmospheric pressure | f. absolute zero |
| g. properties   | h. diffusion            |                  |

## Part A Completion

Use the vocabulary terms in parentheses to complete each of the sentences. Write the complete sentence on the lines provided. Use a dictionary to help you if necessary.

Diffusion theory describes the \_\_\_\_\_, expanding characteristics of gases and demonstrates that the kinetic approach to gases is correct. Gases have great \_\_\_\_\_, \_\_\_\_\_ than the particles of vapours. Gases \_\_\_\_\_, \_\_\_\_\_, and expand when placed in contact with \_\_\_\_\_ matter. The \_\_\_\_\_ characteristics of particles show that the \_\_\_\_\_ particles cannot be pushed into smaller spaces. Gases are not \_\_\_\_\_, and they are \_\_\_\_\_ and expand to fill the \_\_\_\_\_ of their containers. Gases are \_\_\_\_\_ and expand to \_\_\_\_\_ with pressure. \_\_\_\_\_.

1	_____
2	_____
3	_____
4	_____
5	_____
6	_____
7	_____
8	_____
9	_____
10	_____

## Part B True-False

Mark each of the following statements as true (T) or false (F). If you know (FF) or never (FFF).

- \_\_\_\_\_ 11. Gases are compressible.
- \_\_\_\_\_ 12. Gases and vapours have mass.

# Nature Of Gases Section Review Key

**Padhraic Smyth**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Nature Of Gases Section Review Key:**

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