

THE PARTICLE THEORY

The particle theory describes the behaviour of matter. According to this theory, all matter is made up of tiny particles, which are constantly in motion. The behaviour of these particles is influenced by the temperature and pressure of their surroundings.



Fill in the blanks using the word bank provided. Next, draw a diagram to show how particles are arranged in a solid, liquid and gas.

WORD BANK

packed movement shape far vibrate slide closely
freely energetic contact electrical independently

SOLID



In a solid, the particles are tightly _____ together and _____ in place. They have a fixed _____ and volume because the particles are held closely together by strong forces of attraction. The particles can only move in a fixed position and cannot move past one another.

LIQUID

In a liquid, the particles are still close together but have more freedom of _____ than in a solid. They can _____ past one another, allowing liquids to take the shape of their container. However, they still have a fixed volume because the particles are still _____ packed.



Light Particle Model Quiz 3 Answers

William Amestoy



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