# Limiting Reactant Worksheet

Balance the following reaction:

\_\_ NaCl + \_\_ Ba₃(PO₂)₂ → \_\_ Na₃PO₄ + \_\_ BaCl₂

Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

[b] How many grams of excess reactant are left?

Acrylonitrile (C<sub>3</sub>H<sub>3</sub>N) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene (C<sub>3</sub>H<sub>6</sub>) by reacting with nitric oxide (NO) as follows:

What mass of C<sub>3</sub>H<sub>3</sub>N can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

# <u>Limiting Reactant Homework Answers On Chemfiesta</u>

**David Baud** 

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