

Chemistry – Unit 4 Worksheet 4

DMC.5: *I can cite and interpret the evidence that supports the belief that some pure substances are made up of simpler particles (atoms) in a definite ratio.*

Answer the following questions. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

- Two compounds of hydrogen and oxygen are tested. Compound I contains 120.0 g of oxygen and 15.0 g of hydrogen. Compound II contains 32.0 g of oxygen and 2.0 g of hydrogen.
 - Determine the ratio of the mass of oxygen to the mass of hydrogen in each of the compounds.
 - Why are the compounds not the same?
 - What is significant about these mass ratios?
 - If compound I is water, what could be the formula of compound II?
- Nitrogen and oxygen combine to form a variety of compounds. The following data were collected for three different compounds of nitrogen and oxygen:

Analysis Data of Nitrogen & Oxygen Compounds	
Compound	Mass of Nitrogen that combines with every 1.0 g of Oxygen
A	1.750 g
B	0.8750 g
C	0.4375 g

- Additional evidence shows that the formula of compound B is NO. Sketch particle diagrams of molecules of all three compounds.
- Justify your representations above.
- There is no compound between Nitrogen and oxygen with a mass ratio of 2.00 grams of N for every 1.00 grams of oxygen. Why?

Modeling Chemistry Unit 2 Answers

D Kirk



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Chemistry in the Community (ChemCom) American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today's teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards

Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

Towards Sustainable and Scalable Educational Innovations Informed by the Learning Sciences Chee-Kit Looi, David H. Jonassen, Mitsuru Ikeda, 2005 One of the basic principles that underpin the learning sciences is to improve theories of learning through the design of powerful learning environments that can foster meaningful learning Learning sciences researchers prefer to research learning in authentic contexts This book focuses on learning sciences in the Asia Pacific context

Prentice Hall Chemistry, 2000

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Learning Progressions in Science Alicia C. Alonzo, Amelia Wenk Gotwals, 2012-07-30 Learning progressions descriptions of increasingly sophisticated ways of thinking about or understanding a topic National Research Council 2007 represent a promising framework for developing organized curricula and meaningful assessments in science In addition well grounded learning progressions may allow for coherence between cognitive models of how understanding develops in a given domain classroom instruction professional development and classroom and large scale assessments Because of the promise that learning progressions hold for bringing organization and structure to often disconnected views of how to teach and assess science they are rapidly gaining popularity in the science education community However there are significant challenges faced by all engaged in this work In June 2009 science education researchers and practitioners as well as scientists psychometricians and

assessment specialists convened to discuss these challenges as part of the Learning Progressions in Science LeaPS conference The LeaPS conference provided a structured forum for considering design decisions entailed in four aspects of work on learning progressions defining learning progressions developing assessments to elicit student responses relative to learning progressions modeling and interpreting student performance with respect to a learning progressions and using learning progressions to influence standards curricula and teacher education This book presents specific examples of learning progression work and syntheses of ideas from these examples and discussions at the LeaPS conference

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