

Chemistry – Unit 7 Worksheet B

Write balanced chemical equations for the following reactions.

Hydrogen gas (collected in air) is also collected in chlorine (a diatomic gas) and water.

1. Acetylene (C_2H_2) reacts with oxygen (hydrogen) + fluorine for three acetylenes, a fluorine.

$$3\text{C}_2\text{H}_2 + 5\text{F}_2 \rightarrow 6\text{HF} + 3\text{C}_2\text{F}_4$$

2. Calcium carbonate is decomposed (superheating) for three calcium oxide and carbon dioxide.

$$3\text{CaCO}_3 \rightarrow 3\text{CaO} + 3\text{CO}_2$$

3. Barium sulfate reacts with water for three barium hydroxide.

$$3\text{BaSO}_4 + 6\text{H}_2\text{O} \rightarrow 3\text{Ba(OH)}_2 + 3\text{H}_2\text{SO}_4$$

4. Methanethiol (CH_3SCH_3) decomposes to form methane (CH_4) and carbon monoxide.

$$2\text{CH}_3\text{SCH}_3 \rightarrow 2\text{CH}_4 + 2\text{CS}_2$$

5. Zinc reacting with oxygen (a diatomic gas) for three zinc oxide and oxygen.

$$3\text{Zn} + 3\text{O}_2 \rightarrow 3\text{ZnO} + 3\text{O}_2$$

6. Calcium sulfate decomposes (superheating) for three calcium oxide and sulfur dioxide.

$$3\text{CaSO}_4 \rightarrow 3\text{CaO} + 3\text{SO}_2$$

7. Iron reacting with sulfur gas (S_2) for three iron (II) sulfide and hydrogen gas.

$$3\text{Fe} + 3\text{S}_2 \rightarrow 3\text{FeS}_2 + 3\text{H}_2$$

8. A nitrogen-containing carbon compound, $\text{C}_2\text{H}_5\text{N}_3$, decomposes to form nitrogen, C_2H_4 , and nitrogen gas.

$$3\text{C}_2\text{H}_5\text{N}_3 \rightarrow 3\text{C}_2\text{H}_4 + 3\text{N}_2$$

9. Phosphorus, P_4O_{10} , is formed when carbon monoxide reacts with phosphorus gas.

$$3\text{P}_4 + 10\text{CO} \rightarrow \text{P}_4\text{O}_{10} + 10\text{C}$$

Modeling Chemistry Unit 7 Ws3 Answers

Staley Wilbraham (Matta, Waterman)



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Pearson Chemistry Unit 7 Staley Wilbraham (Matta, Waterman),2022

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