

## Chemistry Unit 8 Worksheet 4

### Samples of Every Kind of Problem

*On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.*

1. Calculate the number of moles of potassium chlorate,  $\text{KClO}_3$  (s), that must decompose to produce potassium chloride,  $\text{KCl}$  (s), and 1.8 moles of oxygen gas.

Equation:	$2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$			1.2 moles $\text{KClO}_3$ needed
Before:	1.2	0	0	
Change:	-1.2	+1.2	+1.8	
After:	0	1.2	1.8	

2. In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equation:	$\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$				0.200 moles $\text{HCl}$ needed
Before:	0.100	?	0	0	
Change:	-0.100	-0.200 +	0.100 +	0.100	
After:	0	?	0.100	0.100	

3. Ethane,  $\text{C}_2\text{H}_6$  reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	$2 \text{C}_2\text{H}_6 + 7 \text{O}_2 \rightarrow 4 \text{CO}_2 + 6 \text{H}_2\text{O}$			
Before:	2.20	xs	0	0
Change:	-2.20	-7.70	+4.40	+6.60
After:	0	xs	4.40	6.60

$$\text{O}_2 \text{ needed: } 7.70 \text{ mole} \times \frac{32.0 \text{ g}}{1 \text{ mole}} = 246 \text{ g O}_2 \text{ required}$$

4. Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation:



$$0.73 \text{ g} \times \frac{1 \text{ mole}}{183 \text{ g}} = .0040 \text{ mole}$$

Equation:	$\text{Ni}(\text{NO}_3)_2 + 2 \text{NaOH} \rightarrow \text{Ni}(\text{OH})_2 + 2 \text{NaNO}_3$			
Before:	0.0040	XS	0	0
Change:	-0.0040	-0.0080	+0.0040	+0.0080
After:	0	XS	0.0040	0.0080

$$\text{NaNO}_3 \text{ produced: } 0.0080 \text{ mole} \times \frac{85.0 \text{ g}}{1 \text{ mole}} = 0.68 \text{ g}$$

# Modeling Chemistry Unit 8 Quiz

**K Morrison**



## **Modeling Chemistry Unit 8 Quiz:**

The book delves into Modeling Chemistry Unit 8 Quiz. Modeling Chemistry Unit 8 Quiz is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 8 Quiz, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Unit 8 Quiz
- Chapter 2: Essential Elements of Modeling Chemistry Unit 8 Quiz
- Chapter 3: Modeling Chemistry Unit 8 Quiz in Everyday Life
- Chapter 4: Modeling Chemistry Unit 8 Quiz in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Modeling Chemistry Unit 8 Quiz. The first chapter will explore what Modeling Chemistry Unit 8 Quiz is, why Modeling Chemistry Unit 8 Quiz is vital, and how to effectively learn about Modeling Chemistry Unit 8 Quiz.
3. In chapter 2, the author will delve into the foundational concepts of Modeling Chemistry Unit 8 Quiz. This chapter will elucidate the essential principles that need to be understood to grasp Modeling Chemistry Unit 8 Quiz in its entirety.
4. In chapter 3, this book will examine the practical applications of Modeling Chemistry Unit 8 Quiz in daily life. The third chapter will showcase real-world examples of how Modeling Chemistry Unit 8 Quiz can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Modeling Chemistry Unit 8 Quiz in specific contexts. This chapter will explore how Modeling Chemistry Unit 8 Quiz is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 8 Quiz. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Unit 8 Quiz.

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