

MODELING CHEMISTRY

Initial Question

Exploring chemical and physical changes in a laboratory experiment is not as easy as one may think. The general appearance of a substance can tell us something about events on the molecular level, but this provides limited information. Measuring parameters like conductivity, temperature, pH, or pressure with digital sensors can provide data that helps us understand more about what is going on, but even that may not be enough to completely understand that which is too small to see. This is the challenge of being a chemist.

What tools allow us to examine physical and chemical changes?

Materials and Equipment

Model 1

- Data collection system
- Temperature sensor
- pH sensor
- Conductivity sensor
- Graduated cylinder, 100-mL
- Beakers (6), glass, 100-mL
- Stirring rod
- Unknowns 1A–1D, 100 mL each
- Distilled water wash bottle

Model 2

- Data collection system
- Absolute pressure sensor^{1,2}
- Tubing and tubing connector (2)
- Quick release connector
- Sensor extension cable
- Test tube rack
- Test tubes (3), 20mm × 150 mm, glass
- Rubber stopper, #2, two-hole
- Syringe, 10-mL, to fit stopcock
- Stopcock to fit two-hole stopper
- Graduated cylinder, 10 mL
- Unknown 2A, 2 mL
- Unknown 2B, fill approximately 1/4 of the test tube
- Glycerin, several drops
- Tongs
- Paper towel

Model 3

- Data collection system
- Temperature sensor
- pH sensor
- Conductivity sensor
- Graduated cylinder, 100-mL
- Beaker, glass, 100-mL
- Stirring rod
- Distilled water, 50 mL
- Distilled water wash bottle
- Each group is assigned one of the following:
 - Sucrose ($C_{12}H_{22}O_{11}$), about 0.5 g
 - Sodium chloride ($NaCl$), about 0.5 g
 - Sodium acetate ($NaCH_3COO$), about 0.5 g
 - Calcium (Ca) metal turning, about the size of half a pea
 - Ammonium nitrate (NH_4NO_3), about 0.5 g

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Velavancorresponding



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