

1

The Science of Biology

Chapter Test A

Multiple Choice

Write the letter that best answers the question or completes the statement on the line provided.

- _____ 1. Science differs from other disciplines, such as history and the arts, because science relies on
- facts.
 - testing explanations.
 - observations.
 - theories.
- _____ 2. Measurements made while observing a plant grow 3 cm over a two-week period are called
- inferences.
 - variables.
 - hypotheses.
 - data.
- _____ 3. Based on your observations, you suggest that the presence of water could accelerate the growth of bread mold. This is
- a conclusion.
 - a hypothesis.
 - an experiment.
 - an analysis.
- _____ 4. Suppose that a scientist proposes a hypothesis about how a newly discovered virus affects humans. Other virus researchers would likely
- reject the hypothesis right away.
 - change the hypothesis to fit their own findings.
 - design new experiments to test the proposed hypothesis.
 - assume that the hypothesis is true for all viruses.
- _____ 5. Why is creativity considered a scientific attitude?
- Scientists need creativity to make good posters to explain their ideas.
 - Creativity helps scientists come up with different experiments.
 - Creative scientists imagine the results of experiments without doing them.
 - Scientists who are creative are better at handling and training animals.
- _____ 6. Who reviews articles for peer-reviewed journals?
- friends of the scientists who wrote the articles
 - anonymous and independent experts
 - the scientists who did the experiments
 - people who paid for the experiments
- _____ 7. How does sharing ideas through peer-reviewed articles help advance science?
- Peer-reviewed articles are published only when the ideas they contain have been accepted by most scientists.
 - Experiments in peer-reviewed articles do not need to be repeated.
 - Scientists reading the articles may come up with new questions to study.
 - Ideas in the articles always support and strengthen dominant theories.

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Kaplan



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KY HS Test Prac Wkbks W/Corr Sci 2001 Holt Rinehart & Winston,2002-03 *Kaplan MCAT 528* Kaplan,2015-07-07

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Engineering Production, 1923 *It's about Time* Alane J. Starko, 1986 *Real SAT II* College Entrance Examination Board, 1998-08-04

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Journal of the American Medical Association, 1922 Hoard's Dairyman, 1904

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web jan 5 2016 explanation isotopes are atoms of the same element same atomic number i.e same number of protons in the nucleus but different atomic masses different mass numbers due to different numbers of neutrons in the nucleus for example we may get an isotope of chlorine with 17 protons and 18 neutrons hence atomic mass 35

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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