

# Homework 1: Mixed Ionic/Covalent Compound Naming

*For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.*

1)  $\text{Na}_2\text{CO}_3$

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2)  $\text{P}_2\text{O}_5$

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3)  $\text{NH}_3$

---

4)  $\text{FeSO}_4$

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5)  $\text{SiO}_2$

---

6)  $\text{GaCl}_3$

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7)  $\text{CoBr}_2$

---

8)  $\text{B}_3\text{H}_4$

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9)  $\text{CO}$

---

10)  $\text{P}_4$

---

*For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.*

11) dinitrogen trioxide

---

12) nitrogen

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13) methane

---

14) lithium acetate

---

15) phosphorus trifluoride

---

16) vanadium (V) oxide

---

17) aluminum hydroxide

---

18) zinc sulfide

---

19) silicon tetrafluoride

---

20) silver phosphate

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# Mixed Ionic Compound Homework

**Gareth Pritchard**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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