

**CLASS SET**

**Mixed Stoichiometry Problems**

1. How many moles of  $H_2$  would be required to completely react with  $O_2$  to produce 5 moles of water? **5 mol  $H_2$**
2.  $H_2SO_4 + NaOH \rightarrow Na_2SO_4 + H_2O$ 
  - a. Balance this equation
  - b. What mass of  $H_2SO_4$  would be required to react with 0.75 mol of NaOH? **37g**
3. What mass of  $NO_2$  is formed when NO reacts with 384 g of  $O_2$ ? **1104g**
4. If 30 g of  $H_2SO_4$  reacts with aluminum hydroxide in a double replacement reaction, what mass of water is produced? **11.0g**
5. If 10 L of  $CH_4$  combusts with  $O_2$ , what volume of water will be produced? **20 L  $H_2O$**
6. Tin(II) fluoride,  $SnF_2$ , is used in some toothpastes. It is made by the reaction of tin with hydrogen fluoride according to the following equation.  
 $Sn + 2HF \rightarrow SnF_2 + H_2$   
How many grams of  $SnF_2$  are produced from the reaction of 30 g of HF with Sn? **118 g**
7. In a spacecraft, the carbon dioxide exhaled by astronauts can be removed by its reaction with lithium hydroxide,  $LiOH$ , according to the following chemical equation.  
 $CO_2(g) + 2LiOH(s) \rightarrow Li_2CO_3(s) + H_2O(l)$   
If 448 L of  $CO_2$  is exhaled, the average amount exhaled by a person each day, how much (in grams)  $Li_2CO_3(s)$  is produced? **1480g  $Li_2CO_3$**
8. When 9.8g of aluminum oxide decomposes, how many grams of aluminum metal are produced? **5.2 g Al**
9. How many grams of iodine are produced when 0.72 mol of fluorine react with potassium iodide?
10. How many grams of sodium are required to react with water to produce 5.0g of sodium hydroxide? ( $Na + H_2O \rightarrow NaOH + H_2$ )
11. How many grams of  $H_2$  are produced in the reaction in #10?
12. What volume would be occupied by the resulting hydrogen at STP?
13. 20.0g of K react with  $H_2O$  to produce KOH and  $H_2$  gas. How many liters would the hydrogen gas occupy at STP?
14. If 30.2g of Al react with HCl to produce  $AlCl_3$  and  $H_2$  gas, how many liters of  $H_2$  are produced at STP?
15. How many liters of HCl are produced by the reaction of 5.7L of  $H_2$  with  $Cl_2$ ?

# Mixed Stoichiometry Word Problems And Answers

**Minjie Lin**



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## **Table of Contents Mixed Stoichiometry Word Problems And Answers**

1. Understanding the eBook Mixed Stoichiometry Word Problems And Answers
  - The Rise of Digital Reading Mixed Stoichiometry Word Problems And Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Mixed Stoichiometry Word Problems And Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mixed Stoichiometry Word Problems And Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mixed Stoichiometry Word Problems And Answers
  - Personalized Recommendations
  - Mixed Stoichiometry Word Problems And Answers User Reviews and Ratings
  - Mixed Stoichiometry Word Problems And Answers and Bestseller Lists
5. Accessing Mixed Stoichiometry Word Problems And Answers Free and Paid eBooks
  - Mixed Stoichiometry Word Problems And Answers Public Domain eBooks
  - Mixed Stoichiometry Word Problems And Answers eBook Subscription Services
  - Mixed Stoichiometry Word Problems And Answers Budget-Friendly Options
6. Navigating Mixed Stoichiometry Word Problems And Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Mixed Stoichiometry Word Problems And Answers Compatibility with Devices
  - Mixed Stoichiometry Word Problems And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mixed Stoichiometry Word Problems And Answers
  - Highlighting and Note-Taking Mixed Stoichiometry Word Problems And Answers
  - Interactive Elements Mixed Stoichiometry Word Problems And Answers
8. Staying Engaged with Mixed Stoichiometry Word Problems And Answers

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Mixed Stoichiometry Word Problems And Answers
- 9. Balancing eBooks and Physical Books Mixed Stoichiometry Word Problems And Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mixed Stoichiometry Word Problems And Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mixed Stoichiometry Word Problems And Answers
  - Setting Reading Goals Mixed Stoichiometry Word Problems And Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mixed Stoichiometry Word Problems And Answers
  - Fact-Checking eBook Content of Mixed Stoichiometry Word Problems And Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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