III BOILE Ionic Bonding

Metals and non-metals react together to form _____ compounds. They are held together by ionic bonds. Ionic bonding is a force of ______ attraction between two ____ atoms (called ____). An atom becomes charged by either donating or accepting ___ electrons from another atom. A valence electron is one from the ______ shell of the atom.

Since opposites attract, there is an attraction between the positively charged and negatively charged ions that holds them together.

If an atom ____ an electron, it becomes positively charged. This is then called a ____ always donate electrons. If an atom accepts an electron, it becomes charged. This is called an ______ Non-metals always accept electrons.

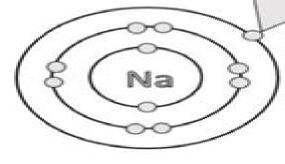
Atoms do this in order to become '______. A stable atom is one _____ of electrons

which has obtained a ______ of electrons.

How do we know which atoms donate and which accept electrons?

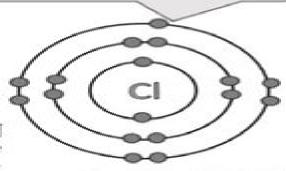
Sodium has ____ lone electron in the outer shell. It is easiest for this atom to ______ one electron than to try and find seven more to fill the outer shell

Chlorine has _____ electrons in its outer shell. Therefore, it is easiest for Chlorine to try and donate seven.



donate electrons and therefore become cations.

> accept electrons and therefore become anions.



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