

# General Chemistry - Unit 1 (Part 1) Review WS - **ANSWERS**

**Directions:** Complete the following questions on a separate piece of paper.

**LT #2: Understand the experiments and the historical theories that lead to the modern atomic theory.**

- Describe Dalton's beliefs about the atom and chemical reactions with atoms.  
*All elements are composed of atoms; all atoms of a given element are identical, that atoms of different elements are different; chemical compounds are formed when atoms combine; atoms are not created or destroyed in chemical reactions (indivisible); solid sphere.*
- Describe the atom according to Thomson's theory.  
*Thomson found that the atom consisted of a positive mass but there were negative pieces scattered throughout the positive area (Plum Pudding Model). He discovered the electron (negatively charged).*
- Describe Rutherford's gold foil experiment. How did he change the model of the atom after the experiment?  
*Rutherford discovered the nucleus contained protons by shooting alpha particles (+) through gold foil. He found that most went through but some bounced back. He found the nucleus, which is very, very small and dense is found in the center of the atom, with the electrons around the outside.*
- What did Chadwick change about the atomic model?  
*Chadwick (along with the Curies) found that the nucleus also contained the neutrons, which have approximately the same mass as the proton, but have no charge.*

**LT #3: Understand how subatomic particles determine the properties of an atom or ion.**

- List the three subatomic particles, their location, their charge and their relative mass.  
*Proton—found in nucleus, + charge, 1 amu (atomic mass unit)  
Neutron—found in nucleus, no charge, 1 amu  
Electron—cloud surrounding the nucleus, - charge, approx. no mass*
- What is a cation and how is it formed?  
*A cation is a positive ion and is formed when an atom loses one or more electrons.*
  - What is an anion and how is it formed?  
*An anion is a negative ion and is formed when an atom gains one or more electrons.*
- Complete the following charts for atoms and ions:

| ATOM          | ISOTOPE SYMBOL                  | ATOMIC # | MASS # | #PROTONS | #ELECTRONS | #NEUTRONS |
|---------------|---------------------------------|----------|--------|----------|------------|-----------|
| Sodium - 23   | <sup>23</sup> <sub>11</sub> Na  | 11       | 23     | 11       | 11         | 12        |
| Aluminum - 27 | <sup>27</sup> <sub>13</sub> Al  | 13       | 27     | 13       | 13         | 14        |
| Chlorine - 35 | <sup>35</sup> <sub>17</sub> Cl  | 17       | 35     | 17       | 17         | 18        |
| Barium - 136  | <sup>136</sup> <sub>56</sub> Ba | 56       | 136    | 56       | 56         | 80        |

| SYMBOL           | MASS | #PROTONS | #ELECTRONS | #NEUTRONS | NET CHARGE |
|------------------|------|----------|------------|-----------|------------|
| K <sup>+1</sup>  | 39   | 19       | 18         | 20        | +1         |
| Br <sup>-1</sup> | 80   | 35       | 36         | 45        | -1         |
| Fe <sup>+3</sup> | 56   | 26       | 23         | 30        | +3         |
| Se <sup>-2</sup> | 79   | 34       | 36         | 45        | -2         |

- Are all atoms of the same element identical? If not, how can they differ?  
*No; isotopes differ by # of neutrons. (Isotopes have the same # of protons and electrons.)*
- Which of lead's four isotopes is most abundant: lead-204, lead-206, lead-207 or lead-208? Explain how you know which isotope is more abundant. *The average atomic mass for lead is 207.2 amu; therefore, the most abundant isotope is lead-207*

**LT #4: Apply the mole concept (molar mass, mole equalities, conversions/dimensional analysis).**

- Calculate the molar mass of each of the following atoms or compounds:
  - zinc  
*65.4 g/mol*
  - barium phosphate, Ba<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>  
*3(137.3) + 2(31.0) + 8(16.0) = 601.9 g/mol*

# Modeling Chemistry U7 Ws 2 V2 Answers

**Huangqi Zhang**



## **Modeling Chemistry U7 Ws 2 V2 Answers:**

**Parallel Computing for Bioinformatics and Computational Biology** Albert Y. Zomaya, 2006-05-24 Discover how to streamline complex bioinformatics applications with parallel computing This publication enables readers to handle more complex bioinformatics applications and larger and richer data sets As the editor clearly shows using powerful parallel computing tools can lead to significant breakthroughs in deciphering genomes understanding genetic disease designing customized drug therapies and understanding evolution A broad range of bioinformatics applications is covered with demonstrations on how each one can be parallelized to improve performance and gain faster rates of computation Current parallel computing techniques and technologies are examined including distributed computing and grid computing Readers are provided with a mixture of algorithms experiments and simulations that provide not only qualitative but also quantitative insights into the dynamic field of bioinformatics *Parallel Computing for Bioinformatics and Computational Biology* is a contributed work that serves as a repository of case studies collectively demonstrating how parallel computing streamlines difficult problems in bioinformatics and produces better results Each of the chapters is authored by an established expert in the field and carefully edited to ensure a consistent approach and high standard throughout the publication The work is organized into five parts Algorithms and models Sequence analysis and microarrays Phylogenetics Protein folding Platforms and enabling technologies Researchers educators and students in the field of bioinformatics will discover how high performance computing can enable them to handle more complex data sets gain deeper insights and make new discoveries

*Combustion* J. Warnatz, Ulrich Maas, Robert W. Dibble, 2013-04-17 Combustion is an old technology which at present provides about 90% of our worldwide energy support Combustion research in the past used fluid mechanics with global heat release by chemical reactions described with thermodynamics assuming infinitely fast reactions This approach was useful for stationary combustion processes but it is not sufficient for transient processes like ignition and quenching or for pollutant formation Yet pollutant formation during combustion of fossil fuels is a central topic and will continue to be so in the future This book provides a detailed and rigorous treatment of the coupling of chemical reactions and fluid flow Also combustion specific topics of chemistry and fluid mechanics are considered and tools described for the simulation of combustion processes For the 3rd edition the text has been thoroughly revised and updated *Combustion* Jürgen Warnatz, Ulrich Maas, Robert W. Dibble, 2012-12-06 Combustion is an old technology which at present provides about 90% of our worldwide energy support Combustion research in the past used fluid mechanics with global heat release by chemical reactions described with thermodynamics assuming infinitely fast reactions This approach was useful for stationary combustion processes but it is not sufficient for transient processes like ignition and quenching or for pollutant formation Yet pollutant formation during combustion of fossil fuels is a central topic and will continue to be so in future This book provides a detailed and rigorous treatment of the coupling of chemical reactions and fluid flow Also combustion specific topics of chemistry and

fluid mechanics are considered and tools described for the simulation of combustion processes For the 2nd edition the parts dealing with experiments spray combustion and soot were thoroughly revised      **Modeling the Fate of Chemicals in the Aquatic Environment** Kenneth L. Dickson, Alan W. Maki, John Cairns, 1982 THIS BOOK IS A RESULT OF THE FOURTH PELLSTON ENVIRONMENTAL WORKSHOP HELD AUGUST 16 21 1981 THIS WORKSHOP AND PREVIOUS MEETINGS HAVE EXAMINED METHODS OF ASSESSING HAZARDS OF CHEMICALS IN THE AQUATIC ENVIRONMENT HAZARD ASSESSMENT IT HAS BEEN DETERMINED REQUIRES AN UNDERSTANDING OF THE RELATIONSHIP BETWEEN THE CONCENTRATION OF CHEMICALS CAUSING AN ADVERSE EFFECT ON AQUATIC LIFE AND ENVIRONMENTAL EXPOSURE CONCENTRATIONS MATHEMATICAL ENVIRONMENTAL FATE MODELS ARE SUGGESTED AS A TOOL FOR PREDICTING ENVIRONMENTAL EXPOSURE CONCENTRATIONS OF CHEMICALS THIS BOOK PRESENTS A STATE OF THE ART OVERVIEW OF THE USE OF THESE MODELS FOR DECISION MAKERS IN ASSESSING THE HAZARDS OF CHEMICALS IN THE AQUATIC ENVIRONMENT AUTHORS SUMMARY ABRIDGED      *Government Reports Announcements & Index* ,1990      Index Medicus ,2002 Vols for 1963 include as pt 2 of the Jan issue Medical subject headings

Valorization of Agri-Food Wastes and By-Products Rajeev Bhat, 2021-08-25 Valorization of Agri Food Wastes and By Products Recent Trends Innovations and Sustainability Challenges addresses the waste and by product valorization of fruits and vegetables beverages nuts and seeds dairy and seafood The book focuses its coverage on bioactive recovery health benefits biofuel production and environment issues as well as recent technological developments surrounding state of the art of food waste management and innovation The book also presents tools for value chain analysis and explores future sustainability challenges In addition the book offers theoretical and experimental information used to investigate different aspects of the valorization of agri food wastes and by products Valorization of Agri Food Wastes and By Products Recent Trends Innovations and Sustainability Challenges will be a great resource for food researchers including those working in food loss or waste agricultural processing and engineering food scientists technologists agricultural engineers and students and professionals working on sustainable food production and effective management of food loss wastes and by products Covers recent trends innovations and sustainability challenges related to food wastes and by products valorization Explores various recovery processes the functionality of targeted bioactive compounds and green processing technologies Presents emerging technologies for the valorization of agri food wastes and by products Highlights potential industrial applications of food wastes and by products to support circular economy concepts      *An Integrated Approach to Software Engineering* Pankaj Jalote, 2012-12-06 A lot has changed in the fast moving area of software engineering since the first edition of this book came out However two particularly dominant trends are clearly discernible focus on software processes and object orientation A lot more attention is now given to software processes because process improvement is considered one of the basic mechanisms for improving quality and productivity And the object oriented approach is considered by many one of the

best hopes for solving some of the problems faced by software developers In this second edition these two trends are clearly highlighted Aseparate chapter has been included entitled Software Processes In addition to talking about the various development process models the chapter discusses other processes in soft ware development and other issues related to processes Object orientation figures in many chapters Object oriented analysis is discussed in the chapter on require ments while there is a complete chapter entitled Object Oriented Design Some aspects of object oriented programming are discussed in the chapter on coding while specific techniques for testing object oriented programs are discussed in the chapter on testing Overall if one wants to develop software using the paradigm of object orientation aB aspects of development that require different handling are discussed Most of the other chapters have also been enhanced in various ways In particular the chapters on requirements specification and testing have been considerably enhanced

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