

Name Solutions

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### Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



| Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|-------|-------------------|-------------------------------------|----------------------------|
| Li    | 1 ve              | transfers 1ve <sup>-</sup> to F     | Li <sup>+</sup>            |
| F     | 7 ve              | accepts 1ve <sup>-</sup> from Li    | F <sup>-</sup>             |

| Reactions                                              | Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|--------------------------------------------------------|-------|-------------------|-------------------------------------|----------------------------|
| 1) $\text{Li} + \text{Cl} \Rightarrow \text{LiCl}$<br> | Li    | 1 ve              | donate 1e <sup>-</sup>              | Li <sup>+</sup>            |
|                                                        | Cl    | 1 ve              | accepts 1e <sup>-</sup>             | Cl <sup>-</sup>            |
| 2) $\text{Ca} + \text{O} \Rightarrow \text{CaO}$<br>   | Ca    | 2 ve              | donates 2e <sup>-</sup>             | Ca <sup>+2</sup>           |
|                                                        | O     | 6 ve              | accepts 2e <sup>-</sup>             | O <sup>-2</sup>            |
| 3) $\text{Be} + \text{F} \Rightarrow \text{BeF}_2$<br> | Be    | 2 ve              | donates 2e <sup>-</sup> to each F   | Be <sup>+2</sup>           |
|                                                        | F     | 7 ve              | each F accepts 1ve each             | F <sup>-</sup>             |
| 4) $\text{Mg} + \text{S} \Rightarrow \text{MgS}$<br>   | Mg    | 2 ve              | donates 2e <sup>-</sup> to S        | Mg <sup>+2</sup>           |
|                                                        | S     | 6 ve              | accepts 2e <sup>-</sup>             | S <sup>-2</sup>            |
| 5) $\text{K} + \text{F} \Rightarrow \text{KF}$<br>     | K     | 1 ve              | donates 1ve <sup>-</sup>            | K <sup>+1</sup>            |
|                                                        | F     | 7 ve              | accepts 1ve <sup>-</sup>            | F <sup>-1</sup>            |

# Ionic Bonding Advanced Chemquest Answers

**Huangqi Zhang**



## **Ionic Bonding Advanced Chemquest Answers:**

## The Enigmatic Realm of **Ionic Bonding Advanced Chemquest Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Ionic Bonding Advanced Chemquest Answers** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of those that partake in its reading experience.

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