Sodium becomes a cation, and chlorine becomes an anion.

Activity A:	Get the Gizmo ready:	100	-
- Control of the Cont	 Click Reset. Check that sodium and chlorine are still selected. 		16

Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and –) attract, while particles with the same charge (+ and + or – and –) repel.

Question: What happens when atoms gain or lose electrons?

- Count: Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms
 in the Gizmo. The first ring holds two electrons, and the second holds eight. (Electrons in the inner rings
 are not shown; you can assume these rings are full.)
 - A. Observe the sodium and chlorine atoms. Assuming that the inner rings are full of electrons, how many electrons are there total in each atom?

Sodium: 11 electrons Chlorine: 17 electrons

B. Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11 protons Chilorine: 17 protons

 Observe: Most atoms are stable with a configuration of eight valence electrons. This is known as the octat rule. How many valence electrons does each atom have?

Sodium: 1 valence electron Chlorine: 7 valence electrons

- Form a bond: Each electron has a charge of 1-, and each proton has a charge of 1+. You can calculate the
 charge of an atom by subtracting the number of electrons from the number of protons. Move an electron
 from the sodium to the chlorine atom.
 - A. What are the charges of each atom now? Sodium: +1 Chlorine: -1

Turn on Show charge to check. These charged atoms are called ions.

B. Is each ion stable? Explain.

Yes, each ion is stable. Chlorine only has 7 valence electrons, so gaining 1 more would lead to a stable full octet. Sodium on the other hand has 1 valence electron, by losing this electron it has just a regular shell of 8 electrons, also becoming a stable octet.

Click Check in the lower right corner of the Gizmo to check.

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Ionic Bonds Gizmo Explore Learning Answers

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Ionic Bonds Gizmo Explore Learning Answers:

Adopting the Song of Expression: An Psychological Symphony within Ionic Bonds Gizmo Explore Learning Answers

In some sort of taken by monitors and the ceaseless chatter of instantaneous transmission, the melodic beauty and emotional symphony produced by the prepared term often fade in to the back ground, eclipsed by the constant noise and disturbances that permeate our lives. Nevertheless, located within the pages of **Ionic Bonds Gizmo Explore Learning Answers** a stunning fictional prize brimming with organic emotions, lies an immersive symphony waiting to be embraced. Crafted by a wonderful composer of language, this interesting masterpiece conducts visitors on a mental trip, well unraveling the hidden tunes and profound impact resonating within each carefully crafted phrase. Within the depths with this moving assessment, we shall investigate the book is key harmonies, analyze its enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

 $\frac{https://www.splashdogs.com/data/publication/default.aspx/Kieso\%20Intermediate\%20Accounting\%2010th\%20Edition\%20Testw20Bank.pdf$

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