

Electron Configuration Practice Worksheet

In the space below, write the unabbreviated electron configurations of the following elements:

- 1) sodium _____
- 2) iron _____
- 3) bromine _____
- 4) barium _____
- 5) neptunium _____

In the space below, write the abbreviated electron configurations of the following elements:

- 6) cobalt _____
- 7) silver _____
- 8) tellurium _____
- 9) radium _____
- 10) lawrencium _____

Determine what elements are denoted by the following electron configurations:

- 11) $1s^2 2s^2 2p^6 3s^2 3p^4$ _____
- 12) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$ _____
- 13) $[\text{Kr}] 5s^2 4d^{10} 5p^3$ _____
- 14) $[\text{Xe}] 6s^2 4f^{14} 5d^6$ _____
- 15) $[\text{Rn}] 7s^2 5f^{11}$ _____

Determine which of the following electron configurations are not valid:

- 16) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$ _____
- 17) $1s^2 2s^2 2p^6 3s^3 3d^5$ _____
- 18) $[\text{Ra}] 7s^2 5f^8$ _____
- 19) $[\text{Kr}] 5s^2 4d^{10} 5p^5$ _____
- 20) $[\text{Xe}]$ _____

Matter And The Periodic Table Writing Electron Configurations Worksheet

Carson-Dellosa Publishing



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the construction of the periodic table based on the ground electronic configurations of 118 chemical elements p 32 **The Periodic Table: A Very Short Introduction** Eric R. Scerri, 2019-07-25 The periodic table of elements first encountered by many of us at school provides an arrangement of the chemical elements ordered by their atomic number electron configuration and recurring chemical properties and divided into periodic trends In this Very Short Introduction Eric R Scerri looks at the trends in properties of elements that led to the construction of the table and shows how the deeper meaning of the table s structure gradually became apparent with the development of atomic theory and in particular quantum mechanics which underlies the behaviour of all of the elements and their compounds This new edition publishing in the International Year of the Periodic Table celebrates the completion of the seventh period of the table with the ratification and naming of elements 113 115 117 and 118 as nihonium moscovium tennessine and oganesson Eric R Scerri also incorporates new material on recent advances in our understanding of the origin of the elements as well as developments concerning group three of the periodic table ABOUT THE SERIES The Very Short Introductions series from Oxford University Press contains hundreds of titles in almost every subject area These pocket sized books are the perfect way to get ahead in a new subject quickly Our expert authors combine facts analysis perspective new ideas and enthusiasm to make interesting and challenging topics highly readable *Atoms, Molecules & Elements Gr. 5-8* George Graybill, 2007-09-01 Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource makes the periodic table easier to understand Begin by answering what are atoms See how the atomic model is made up of electrons protons and neutrons Find out what a molecule is and how they differ from elements Then move on to compounds Find the elements that make up different compounds Get comfortable with the periodic table by recognizing each element as part of a group Examine how patterns in the period table dictate how those elements react with others Finally explore the three important kinds of elements metals nonmetals and inert gases Aligned to the Next Generation Science Standards and written to Bloom s Taxonomy and STEAM initiatives additional hands on experiments crossword word search comprehension quiz and answer key are also included

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 Electron shell List of elements with electrons per shell Periodic table Layout Rutherfordium Chemical Antimony Group 8
 element Lanthanum Chemical properties Extended periodic table Eka superactinides Manganese Promethium Physical
 properties and much more **The Basics of Atoms and Molecules** Tom Jackson, 2013-12-15 A fun filled introduction to
 matter the elements of the periodic table atoms electrons reactions and bonding and radioactivity this volume provides young
 adults with chemistry examples that reflect their real world interconnections in science Key terms easy experiments and
 clear illustrations help to guide students through chemical applications A chapter about Niels Bohr and his model for the
 atom honors his contribution to the understanding of atomic structure and to nuclear fission Tools and techniques such as a
 scanning tunneling microscope Rutherford's gold foil experiment and a mass spectrometer highlight this instructive text that
 is aligned to the Common Core Standards **Chemistry** David John Cole-Hamilton, Iain Patterson, 2001 **The Periodic**
Table of Elements .paul F. Kisak, 2015-10-27 The periodic table is a tabular arrangement of the chemical elements ordered
 by their atomic number number of protons in the nucleus electron configurations and recurring chemical properties The
 table also shows four rectangular blocks s p d and f block In general within one row period the elements are metals on the
 lefthand side and non metals on the righthand side The rows of the table are called periods the columns are called groups Six
 groups columns have names as well as numbers for example group 17 elements are the halogens and group 18 the noble
 gases The periodic table can be used to derive relationships between the properties of the elements and predict the
 properties of new elements yet to be discovered or synthesized The periodic table provides a useful framework for analyzing
 chemical behavior and is widely used in chemistry and other sciences This book presents a thorough study of The Periodic
 Table of The Elements **Periodic Table Basic** Speedy Publishing, 2014-09-04 Chemistry classes can be some of the most
 difficult classes for students There are many formulas numbers and calculations to be done and memorized Students are
 responsible for many tasks during the school year Studying the periodic table of the elements of chemistry and physics can
 be overwhelming A periodic table study guide can help students remember the chemical numbers and atomic weights of
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