

## BUILD Vocabulary

**ionic bond** Chemical bond formed when one or more electrons are transferred from one atom to another.

**ion** atom that has a positive or negative charge.

**covalent bond** type of bond between atoms in which the electrons are shared.

**molecule** smallest unit of most compounds that displays all the properties of that compound.

**van der Waals forces** weak attraction that develops between oppositely charged regions of nearby molecules.

**Preface** The prefix *co-* and *co-* mean "together." A compound is made up of elements joined together, and a covalent bond connects atoms that share electrons together. **How are covalent bonds different from ionic bonds?**

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## Chemical Bonds

**KEY QUESTION** What are the main types of chemical bonds?

The atoms in compounds are held together by chemical bonds. Bonds are formed by the electrons in the outer shell, which are called valence electrons. The main types of chemical bonds are ionic bonds and covalent bonds.

**Ionic Bonds** **Ionic bonds** are formed when one or more electrons are transferred from one atom to another. This transfer causes the atom losing the electrons to become positively charged. The atom gaining the electrons becomes negatively charged. The positively and negatively charged atoms become **ions**. The oppositely charged ions are attracted to each other and form an ionic bond.

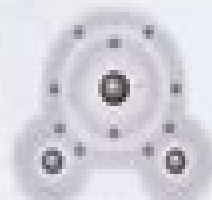
**Covalent Bonds** Sometimes atoms share electrons rather than transferring them. These shared electrons move around the nuclei of both atoms. The bond formed by sharing electrons is called a **covalent bond**. Atoms can share between two and six electrons. The structure that results is known as a **molecule**. A molecule is the smallest unit of most compounds.

**Weak Interactions** The strongest chemical bonds are ionic or covalent. Weak interactions between atoms are also important. Within living organisms, many molecules interact only briefly. **Van der Waals forces** are one type of weak interaction between molecules. They occur when molecules are very close together.

Hydrogen bonds are another type of weak interaction. They occur between a hydrogen atom of one molecule and an oxygen or nitrogen atom of a neighboring molecule.

### Visual Reading Skill Covalent Bonds

1. What kind of chemical bond holds the water molecule together? \_\_\_\_\_
2. What particles are shared between the atoms in the molecule? \_\_\_\_\_
3. What is the other main type of chemical bond? \_\_\_\_\_
4. What is the difference between covalent and ionic bonds? \_\_\_\_\_



Water molecule ( $H_2O$ )

## Properties of Water

**READING TOOL** Cause and Effect As you read the part of the lesson under "The Water Molecule," use the table below to list the causes and effects of the properties of water. Many of the causes may be the same, but each characteristic will have a different effect. Part of the table is completed for you.

Characteristic of Water	Cause	Effect
Hydrogen Bonding		The slightly negative end of a water molecule is attracted to the slightly positive end of another water molecule.
Expands Upon Freezing		
Dissolves Many Substances	Water forms hydrogen bonds.	
Cohesion		
Adhesion		
Heat Capacity		

# Miller And Levine Biology Chapter 17 Study Guide

**Léo Pomar**



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Sandra A. MacDonald, Kathleen Deska Pagana, Timothy J. Pagana, Theresa Noel Pagana, 2022-10-21 Written exclusively for a Canadian market Pagana's Canadian Manual of Diagnostic and Laboratory Tests 3rd Edition provides clear concise coverage of more than 700 of the most commonly performed tests with Canadian lab values SI units Canadian cultural considerations and unique Canadian content Its many user friendly features include an easy to understand writing style full colour illustrations and a logical organization Each test entry is presented in a consistent format to provide quick access to information on specimen collection normal findings indications test explanation procedure and patient care and test results and clinical significance as well as any applicable contraindications potential complications critical values interfering factors and related tests This edition has been updated to reflect the latest procedures equipment and techniques along with 35 of the most current laboratory and diagnostic tests Plus Canadian guidelines are highlighted by a maple leaf icon for easy reference UNIQUE Canadian guidelines are highlighted by a maple leaf icon in the text margin for easy identification and reference UNIQUE A description of commonly performed laboratory methods explains methods used to evaluate blood urine spinal fluid and other specimens UNIQUE Cultural Considerations boxes highlight important aspects of working with patients from the diverse cultural and racial backgrounds of the Canadian population such as Indigenous communities UNIQUE Coverage of the clinical significance of test results explains why a given test result indicates specific diseases UNIQUE SI units in the Normal Findings section of appropriate tests offer quick and easy reference conventional units also included UNIQUE Related Tests sections list tests that provide similar information or are used to evaluate the same body system disease process or symptom Information on Canadian standard precautions and procedures covers those set forth by the Canadian Nuclear Safety Commission as well as Canadian screening protocols such as those set out in the Canadian Cancer Society Screening Guidelines Blood studies coverage includes Canadian test tube colouring classifications and guidelines for the correct order and process of collecting blood samples in Canada Necessary consideration is given to the differences between provinces territories and institutions in regard to privacy legislation obtaining consent agency guidelines procedure protocols and the availability of tests Coverage of Canadian privacy laws and legislation includes PHIPA and PIPEDA the Canadian Labour Code policies for DNA collection reporting of infections such as Chlamydia and much more Up to date Canadian statistics are provided for topics such as STDs and C difficile     **Biofilms in Infection Prevention and Control**  
Steven L. Percival, David Williams, Tracey Cooper, Jacqueline Randle, 2014-01-30 Biofilms in Infection and Disease Control A Healthcare Handbook outlines the scientific evidence and rationale for the prevention of infection the role biofilms play in infection control and the issues concerning their resistance to antimicrobials This book provides practical guidance for

healthcare and infection control professionals as well as students for preventing and controlling infection Biofilms are the most common mode of bacterial growth in nature Highly resistant to antibiotics and antimicrobials biofilms are the source of more than 65 percent of health care associated infections HCAI which according to the WHO affect 1 4 million people annually Biofilms are involved in 80 percent of all microbial infections in the body including those associated with medical devices such as catheters endotracheal tubes joint prostheses and heart valves Biofilms are also the principle causes of infections of the middle ear dental caries gingivitis prostatitis and cystic fibrosis Importantly biofilms also significantly delay wound healing and reduce antimicrobial efficiency in at risk or infected skin wounds Provides specific procedures for controlling and preventing infection Includes case studies of HCAI and identifies appropriate treatments Presents national government standards for infection prevention and control Includes extensive references and links to websites for further information

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**Forthcoming Books** Rose Arny,2003      [Human Nutrition - E-Book](#) Catherine Geissler,Hilary Powers,2010-10-04 This title is now available under ISBN 9780702044632 This 12th edition of Human Nutrition has been fully updated by a renowned team of international experts to ensure to ensure authoritative content and a global perspective It provides a comprehensive resource for all those in the field of nutrition and other health sciences Comprehensive coverage of nutrition in one concise volume with additional material and interactive exercises on website A similar logical chapter structure throughout and textbook features in each chapter learning objectives key point summaries and text boxes facilitate learning and revision Incorporates latest research for example on organic foods and sustainable agriculture Team of contributors of international repute from 11 countries guarantees authoritative text New chapter on dietary reference values N New section on electrolytes and water balance Expanded section on HIV Website updating between editions online only chapters on food commodities e g cereals vegetables and fruit meat fish egg milk and milk products online examples of calculations and interactive exercises

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