

# LAW EXAM CHEAT SHEET

## JUDGES MAKING LAW

★ First Essay Question

### EXAMPLE QUESTIONS

1. Do judges make the law? Should they? Do they have more freedom to make law when applying the common law, compared with statutes? How is their power to make law constrained?
2. "Judges in New Zealand have a remarkable amount of discretion to make law; this is inconsistent with the democracy and the rule of law "

Analyse the extent to which judges have the powers to make law, any limits on that power, and the problems associated with it. Refer to cases studied in the course where relevant.

3. In New Zealand, do judges have unchecked power to make law according to **judicial discretion**?
4. Discuss how the common law, equity, statute law and judicial making is relevant to trade mark law.

## HOW JUDGES MAKE LAW

### COMMON LAW

- Have freedom to develop the law in response to the changing circumstances of society.
- Can set new doctrines of precedents when needed.
- Invent foundational things under common law.
  - ◆ E.g. causes of action, negligence and defamation.
- Example: the tort of 'intrusion into seclusion'
  - ◆ *C v Holland* 2012 saw the high court invent this tort.
  - ◆ Entitled to damages when your reasonable expectation of privacy is breached, e.g. sitting in your backyard and having some take and publish pictures of you.

### PASSING OFF

- The tort of passing off is part of trade mark law.
- It developed to protect brands.
- Common law developed this tort to prevent various traders from taking advantage of someone else's brand, and 'passing it off' as their own.
- The tort of passing off was set as a precedent in the case of *Perry v Truefit*

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over  
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point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like  
a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise  
calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of  
the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube  
end c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the melting point apparatus compound a had a melting point range of 114.6-115.5 c thus revealing it to be acetanilide compound b had a melting point range of 122.6-124.5 c thus

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measured pressure a compound's normal boiling point refers to its boiling point at a pressure of 760 mm Hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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laboratory partners samantha and mackenzie goals ☐ correctly determine the boiling point of an unknown liquid ☐ measure the refractive index of an unknown ☐ use the gas chromatogram to determine the mass percentage of a mixture and

*boiling point determination faculty of science* - Jun 18 2023

50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

*chem1212 boiling point lab report course hero* - Aug 08 2022

introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

6.2b step by step procedures for boiling point determination - Mar 15 2023

there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a Thiele tube the most straightforward method uses a Thiele tube and has the advantage of using less than 0.5 mL of material