

Name: \_\_\_\_\_

**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 1) What is the solvent in an aqueous solution?

**ESSAY.** Write your answer in the space provided or on a separate sheet of paper.

- 2) Explain why water is a good solvent for ionic substances.

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 3) Of the species below, only \_\_\_\_\_ is not an electrolyte.  
 A) HCl                      B) Ar                      C) NaCl                      D) KOH                      E)  $\text{Rb}_2\text{SO}_4$
- 4) What is the molarity of an aqueous solution containing 75.3 g of glucose ( $\text{C}_6\text{H}_{12}\text{O}_6$ ) in 35.5 mL of solution?  
 A) 0.197                      B) 2.12                      C) 3.52                      D) 1.85                      E) 11.8
- 5) The concentration of species in 500 mL of a 2.104 M solution of sodium sulfate is \_\_\_\_\_ M sodium ion and \_\_\_\_\_ M sulfate ion.  
 A) 4.208, 2.104                      B) 1.052, 1.052                      C) 2.104, 1.052                      D) 2.104, 4.208                      E) 2.104, 2.104
- 6) What is the molarity of a solution prepared by diluting 43.72 mL of 1.005 M aqueous  $\text{K}_2\text{Cr}_2\text{O}_7$  to 500 mL?  
 A) 0.870                      B) 87.9                      C) 0.0879                      D) 0.0115                      E) 0.0218
- 7) When aqueous solutions of \_\_\_\_\_ are mixed, a precipitate forms.  
 A) NaI and KBr  
 B)  $\text{Li}_2\text{CO}_3$  and CsI  
 C)  $\text{K}_2\text{SO}_4$  and  $\text{CrCl}_3$   
 D)  $\text{NiBr}_2$  and  $\text{AgNO}_3$   
 E) KOH and  $\text{Ba}(\text{NO}_3)_2$
- 8) What are the spectator ions in the reaction between KOH (aq) and  $\text{HNO}_3$  (aq)?  
 A)  $\text{H}^+$  and  $\text{NO}_3^-$   
 B)  $\text{OH}^-$  only  
 C)  $\text{K}^+$  and  $\text{NO}_3^-$   
 D)  $\text{K}^+$  and  $\text{H}^+$   
 E)  $\text{H}^+$  and  $\text{OH}^-$

**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 9) What is the concentration of HCl in a 0.100 M solution of HCl? Why?

# Multiple Choice Ap Chemistry Practice Test 1

**Richard Bailey**



## **Multiple Choice Ap Chemistry Practice Test 1:**

The book delves into Multiple Choice Ap Chemistry Practice Test 1. Multiple Choice Ap Chemistry Practice Test 1 is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Multiple Choice Ap Chemistry Practice Test 1, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Multiple Choice Ap Chemistry Practice Test 1
- Chapter 2: Essential Elements of Multiple Choice Ap Chemistry Practice Test 1
- Chapter 3: Multiple Choice Ap Chemistry Practice Test 1 in Everyday Life
- Chapter 4: Multiple Choice Ap Chemistry Practice Test 1 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Multiple Choice Ap Chemistry Practice Test 1. This chapter will explore what Multiple Choice Ap Chemistry Practice Test 1 is, why Multiple Choice Ap Chemistry Practice Test 1 is vital, and how to effectively learn about Multiple Choice Ap Chemistry Practice Test 1.

3. In chapter 2, the author will delve into the foundational concepts of Multiple Choice Ap Chemistry Practice Test 1. The second chapter will elucidate the essential principles that must be understood to grasp Multiple Choice Ap Chemistry Practice Test 1 in its entirety.

4. In chapter 3, the author will examine the practical applications of Multiple Choice Ap Chemistry Practice Test 1 in daily life. The third chapter will showcase real-world examples of how Multiple Choice Ap Chemistry Practice Test 1 can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Multiple Choice Ap Chemistry Practice Test 1 in specific contexts. This chapter will explore how Multiple Choice Ap Chemistry Practice Test 1 is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Multiple Choice Ap Chemistry Practice Test 1. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Multiple Choice Ap Chemistry Practice Test 1.

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