

Gravimetric Analysis Questions

Q-1: Which of the following crucibles is used when precipitates are dried in the muffle furnace in gravimetric analysis?

a) Crucible made of porcelain

b) Crucible made of silica

c) Both of the above

d) None of the above

Answer: c) Both of the above

Explanation: Crucible made of porcelain and silica are used when precipitates are dried in the muffle furnace in gravimetric analysis.

Q-2: Which of the following gravimetric analysis uses temperature as a function to measure a material's physical and chemical properties?

a) Electrogravimetry

b) Thermogravimetry

c) Volatilisation gravimetry

d) Precipitation gravimetry

Answer: b) Thermogravimetry is a thermal analysis method that measures changes in physical and chemical properties of materials as a function of increasing temperature or time.

Q-3: What is the principle of gravimetric analysis?

Answer: The gravimetric analysis principle is based on determining the mass % of an ion in a known amount of impure compound. It is then used to determine the mass % of the same ion in a known amount of impure substance.

Q-4: What are the three theories that govern gravimetry?

Answer: The law of mass action and reversible reactions, the principle of solubility product, and the common ion effect are the fundamental principles and theories of gravimetric analysis.

Q-5: What is the distinction between gravimetric and volumetric analysis?

Answer: Gravimetric and volumetric analysis are both quantitative methods for calculating the amount of sample in a solution or the purity of a compound; gravity analysis determines the mass of the analyte, while volumetric analysis determines the volume of the analyte.

Q-6: What do you mean by precipitation gravimetry?

Answer: Precipitation gravimetry is an analytical technique that uses a precipitation reaction to separate the ions in a solution. The precipitant, also known as the precipitating agent, is the chemical

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