



Subject Title: PHARMACEUTICS-II

WINTER- 19 EXAMINATION

Subject Code:

0811

1	c)	<p><b>Translate the following Latin terms in English.</b></p> <p><b>Jantaculum</b> -Breakfast</p> <p><b>Capiendus</b>- To be taken</p> <p><b>Haustous</b>: A droughts</p> <p><b>Hora Somni</b>: at bed time or before sleep.</p>	(2M = 0.5 X 4)								
1	d)	<p><b>Why most of emulsion appears white or opaque.</b></p> <p>Emulsions usually appear cloudy or white because light is scattered off the phase interphases between the components in the mixture. If all of the light is scattered equally, the emulsion will appear white.</p>	2M								
1	e)	<p><b>Give any four properties of suppositories base.</b></p> <ul style="list-style-type: none"><li>● It must retain the shape and size.</li><li>● It should melt at body temperature.</li><li>● It should shrink sufficiently to remove from mould.</li><li>● It should permit incorporation of drug.</li><li>● It should be physically stable on storage.</li><li>● It should not be soften or harden on storage.</li><li>● It should be compatible with variety of drugs.</li><li>● It should not interfere in release or absorption of drug.</li><li>● It should be non-irritant.</li></ul>	(2M = 0.5 X 4)								
1	f)	<p><b>How will you dispense a powder containing eutectic mixture?</b></p> <p>When two or more substances are mixed together they liquefy due to the formation of new compound which has a lower melting point than room temperature such substances are called as eutectic mixtures.</p> <p>The can be dispensed in separate packets or equal quantity of inert solid is mixed.</p> <p>Rx</p> <table><tr><td>Menthol</td><td>5 parts</td></tr><tr><td>Camphor</td><td>5 parts</td></tr><tr><td>Ammonium chloride</td><td>30 parts</td></tr><tr><td>Magnesium carbonate</td><td>60 parts</td></tr></table>	Menthol	5 parts	Camphor	5 parts	Ammonium chloride	30 parts	Magnesium carbonate	60 parts	2M
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# Msbte Answer Paper Pharmacy 2 Year

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