

JEE Main 2020 Paper



Sol. 2



P - OH bonds = 4

P = O bonds = 2

P - O - P linkage = 1

Ans. 4, 2, 1

21. The mole fraction of glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) in an aqueous binary solution is 0.1. The mass percentage of water in it, to the nearest integer, is _____.

Sol. 47 %

$$x_{\text{Glucose}} = 0.1$$

$$\begin{aligned} \text{mass\% of glucose} &= \frac{0.1 \times 180}{0.1 \times 180 + 0.9 \times 18} \times 100 \\ &= \frac{1800}{18 + 16.2} \\ &= \frac{1800}{34.2} \\ &= 52.63\% \\ &= 53\% \end{aligned}$$

$$\therefore \text{mass \% of H}_2\text{O} = 47\%$$

22. The volume strength of 8.9 M H_2O_2 solution calculated at 273 K and 1 atm is _____. ($R = 0.0821 \text{ L atm K}^{-1} \text{ mol}^{-1}$) (rounded off to the nearest integer)

Sol. 100

$$\begin{aligned} \text{Vol. strength} &= \frac{8.9}{2} \times \frac{0.0821 \times 273}{1} \\ &= 99.73 \\ &= 100 \end{aligned}$$

23. An element with molar mass $2.7 \times 10^{-2} \text{ kg mol}^{-1}$ forms a cubic unit cell with edge length 405 pm. If its density is $2.7 \times 10^3 \text{ kg m}^{-3}$, the radius of the element is approximately _____ $\times 10^{-12} \text{ m}$ (to the nearest integer).

Sol. 143

$$\text{Density} = \frac{Z \times \text{GMM}}{N_A \times a^3}$$

$$\begin{aligned} 2.7 \times 10^3 &= \frac{Z \times 2.7 \times 10^{-2}}{6.023 \times 10^{23} \times (405 \times 10^{-12})^3} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{23-36+3+2} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{-8} \\ Z &= 4 \end{aligned}$$

FCC

$$4R = \sqrt{2} \times a$$

$$R = \frac{405}{2\sqrt{2}} \times 10^{-12} = 143.21 \times 10^{-12} \text{ m}$$

= 143 ans

Jee Mains 2question Paper Solution By Fiit Jeee

Mamta Batra



Jee Mains 2question Paper Solution By Fiit Jee:

JEE Main 13 Years Question Paper Prepjee, 2014-08-01 THIS BOOK HAS BEEN DEVELOPED AND DESIGNED TO OVERCOME WITH PRESSING NEED OF JEE MAIN ASPIRANTS THE BOOK CONTAINS PAPERS OF AIEEE JEE MAIN OF PAST 13 YEARS WITH ANSWER KEY AND DETAILED SOLUTION FOR THREE YEARS PAPERS THIS WILL GUIDE THE COURSE OF PREPARATION OF AN ASPIRANT IN A DEFINITE AND REQUIRED DIRECTION LIKE RADAR MOREOVER SOLUTION OF GIVEN YEARS i.e. 2012 2014 IS IN APPROPRIATE MANNER SO THAT AVERAGE STUDENT CAN UNDERSTAND AND SOLVE PROBLEMS EASILY THIS BOOK ALSO HELPS STUDENTS IN CONCEPT BUILDING **16**

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